

Answers to Even-Numbered and Challenge Questions and Problems

Appendix 5

Chapter 1

2. (a) element
 (b) mixture
 (c) compound
 (d) mixture

4. (a) heterogeneous mixture
 (b) heterogeneous mixture
 (c) solution

6. (a) chromatography
 (b) distillation or chromatography

8. (a) Cu (b) C (c) Br (d) Al

10. (a) manganese (b) gold (c) lead (d) silver

12. (a) yardstick (b) buret/pipet (c) thermometer

14. 177°C ; $4.50 \times 10^3\text{ K}$

16. 35°C ; 95°F

18. (a) 3 (b) 4 (c) 4 (d) 1 (e) 5

20. (a) 17.25 cm (b) 169 lb
 (c) $5.00 \times 10^2\text{ ^{\circ}\text{C}}$ (d) 198 oz

22. (a) $4.0206 \times 10^3\text{ mL}$
 (b) 1.006 g
 (c) $1.001 \times 10^2\text{ ^{\circ}\text{C}}$

24. b and c

26. 5 is exact; 4000 is ambiguous; 17 has two and 18.5 has three.

28. (a) 10.6 (b) 80.1 (c) 0.40 (d) 0.0923
 (e) 1.4×10^{-2}

30. 0.04 m^3

32. (a) less than (b) greater than (c) equal

34. (a) 215.6 mL (b) 13.16 m^3 (c) 0.2278 qt

36. (a) 1.15078 mi (b) 1852 m (c) 25 mph

38. (a) USD 3.09 (b) USD 4.5 (c) $2.1 \times 10^3\text{ mi}$

40. $1.85 \times 10^{-3}\text{ g/mL}$

42. $3.97 \times 10^{-2}\text{ L}$

44. 393.8 mg

46. 1.07 g/mL

48. 1.6 g/mL

50. 508 m

52. $3.88 \times 10^4\text{ g}$; 85.4 lb

54. supersaturated; 5 g

56. (a) homogeneous; 4.7 g more (b) 9.1 g

58. 479 rems; $1.6 \times 10^5\text{ mammograms}$

60. 320.0 g

62. $6.3 \times 10^{-6}\text{ m}$

64. 8.2 g/mL

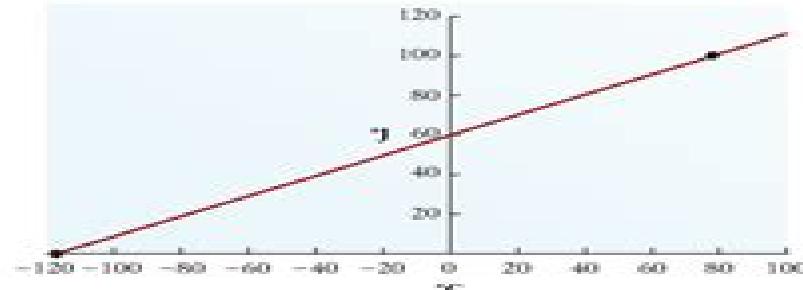
66. (a) I (b) I (c) I (d) E (e) E

68. See pages 16–19

70. 3

72. (a) A (b) -2.3°C (c) no; (see answer to (a))

73. (a)



(b) 0.51
 (c) 60
 (d) $^{\circ}\text{F} = 0.512\text{ (}^{\circ}\text{C}\text{)} + 60$

74. $320^{\circ}\text{F} = 160^{\circ}\text{C}$

75. 1.2 km^2

76. 21.9 cm

77. $8.1 \times 10^{-3}\text{ g Pb}$

78. 6.87 g/cm^3

Chapter 2

2. See page 24

4. (a) none (b) conservation of mass
 (c) none

6. Rutherford; see pages 25–26

8. (a) 86 p^- (b) 1.34 m

10. (a) ^{56}Fe ; ^{58}Fe
 (b) They differ in the number of neutrons. Fe-54 has 28, whereas Fe-56 has 30.

12. (a) 95 (b) 241 (c) ^{241}Am

14. (a) A is arsenic: 33p^- , $42n$, $33e^-$
 (b) L is vanadium: 23p^- , $28n$, $23e^-$
 (c) Z is xenon: 54p^- , $77n$, $54e^-$

16. (a) isobars: Cr-54 and Fe-54 ; Fe-58 and Ni-58 isotopes; Fe-54 and Fe-58
 (b) number of protons
 (c) Cr-54 and Ni-58

18. (a) < (c) < (b) < (d)

20. O-16

22. b

24. 20.2 amu

26. 64.94 amu; ^{63}Cu

28. lightest isotope 78.6%; heaviest isotope 11.4%

30. (a) two: HCl-35 and HCl-37 (b) 36 and 38

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