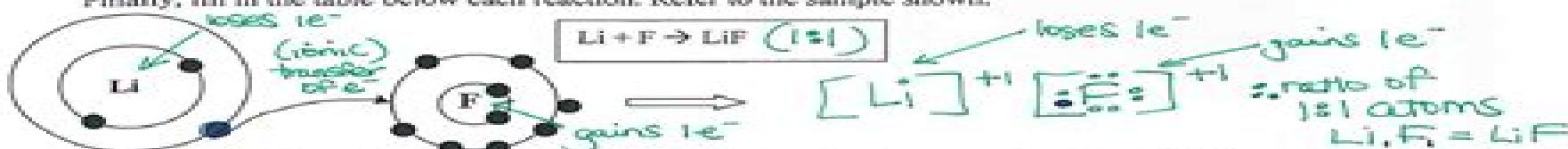


Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1ve		Li ⁺
F	7ve	transfers 1ve to F accepts 1ve from Li	F ⁻

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) Li + Cl \rightarrow LiCl	Li	1ve	donates 1e ⁻	Li ⁺
	Cl	7ve	accepts 1e ⁻	Cl ⁻
2) Ca + O \rightarrow CaO	Ca	2ve	donates 2e ⁻	Ca ⁺
	O	6ve	accepts 2e ⁻	O ²⁻
3) Be + F \rightarrow BeF ₂	Be	2ve	donates 2e ⁻ to each F.	Be ⁺
	F	7ve	each F accepts 1ve each	F ⁻
4) Mg + S \rightarrow MgS	Mg	2ve	donates 2e ⁻ to S	Mg ⁺
	S	6ve	accepts 2e ⁻	S ²⁻
5) K + F \rightarrow KF	K	1ve	donates 1e ⁻	K ⁺
	F	7ve	accepts 1e ⁻	F ⁻

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