

Density Worksheet
Physical Science
D=m/V

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the **density** of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher **density**.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - A. mass = 453 g; volume = 225 cm³
 - B. mass = 5.0 g; volume = 10.0 cm³
 - C. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Answers To Aventa Physical Science

Joacim Rocklöv

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Physical Sciences ,2013 **Physical Science Made Easy** Theodore G. Phillips,William Haskel Colburn,1950

Physical Science National Learning Corporation,2018 The DSST Physical Science Passbook R prepares candidates for the DSST exam which enables schools to award credit for knowledge acquired outside the normal classroom environment It provides a series of informational texts as well as hundreds of questions and answers in the areas that will likely be covered on your upcoming exam including but not limited to physics electricity and magnetism Glossy chemical reactions atomic structure and more Answers Introduction to Physical Science World O F Atoms McCue,1963-05-01 **Physics** Physical Science Study Committee,Uri Haber-Schaim,1976 **Experiences in Physical Science** Laidlaw Brothers,1983 An Approach to Physical Science Physical Science for Nonscience Students Project,1969 Physical Science Mark A. Carle,1991 **Physical Science Workbook** ,1995 Senior Physical Science ,1969 A First Course in Physical Science

Robert P. Bauman,1987 This text provides an understanding of the foundations and structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each topic

Physical Science Jerry Schad,1996 Schad successfully leads students on an exploration of key physical concepts and their applications in the four disciplines of physical science By repeatedly linking the fundamental discoveries and ideas of physics and their applications in other fields the author helps students see the underlying simplicity and unity of nature and presents a balanced coverage of topics Schad writes conceptually and descriptively introducing mathematical formula and sample problems with answers with discretion Features include global surveys boxes to introduce students to the frontiers of science and original art and diagrams **Physical Sciences** Santie Du Plessis,2022 **Principles of Physical Science** Francis T.

Bonner,Melba Phillips,Jane Raymond,1971 The Best of the Lastest Exam Questions & Answers in Physical Science HG Anne Line,Peter S. Westwood,Stephen Whitehouse,2002 **Physical Science** Robert H. Marshall,Donald H. Jacobs,Media Materials, Inc,1987

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