

7.1

IONS

Section Review

Objectives

- 1. Understand the structure of ionic compounds using periodic table trends
- 2. Importance of crystal
- 3. Molecular formula versus
- 4. Empirical formula

Vocabulary

- 1. ionic compound
- 2. crystal lattice structure
- 3. formula unit
- 4. crystal unit

Part A Completion

Use the information on ions to determine the formula of the compound. Write the formula of the compound in the space provided.

1. A metal with a valence of +2 and a nonmetal with a valence of -3 combine to form a compound. Write the formula of the compound.

2. A metal with a valence of +1 and a nonmetal with a valence of -2 combine to form a compound. Write the formula of the compound.

3. A metal with a valence of +2 and a nonmetal with a valence of -3 combine to form a compound. Write the formula of the compound.

4. A metal with a valence of +1 and a nonmetal with a valence of -2 combine to form a compound. Write the formula of the compound.

5. A metal with a valence of +2 and a nonmetal with a valence of -3 combine to form a compound. Write the formula of the compound.

6. A metal with a valence of +1 and a nonmetal with a valence of -2 combine to form a compound. Write the formula of the compound.

7. A metal with a valence of +2 and a nonmetal with a valence of -3 combine to form a compound. Write the formula of the compound.

8. A metal with a valence of +1 and a nonmetal with a valence of -2 combine to form a compound. Write the formula of the compound.

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RS Peters



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