

Name _____

Date _____

Reading the Periodic Table KEY1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 2.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 2 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group (family) do these belong?

a. Cobalt transition metals

d. Silicon group (I) nonmetals

b. Sodium alkali metals

e. Chlorine halogens

c. Argon noble gases

f. Magnesium alkaline earth metals

metals

8. Find iron on the periodic table:

a. What is the atomic number of iron? 26

b. To which group does iron belong? Transition metals

9. Name the elements that have properties similar to that of chlorine.

fluorine, bromine, iodine, chlorine, bromine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
silicon	Si	14	3	14
nitrogen	N	7	2	15
krypton	Kr	36	4	18
gold	Au	79	6	19
chromium	Cr	24	4	18
chlorine	Cl	17	3	17

2 Exploring The Periodic Table Answer Key

Lars O. Werme

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Table of Contents 2 Exploring The Periodic Table Answer Key

1. Understanding the eBook **2 Exploring The Periodic Table Answer Key**
 - The Rise of Digital Reading **2 Exploring The Periodic Table Answer Key**
 - Advantages of eBooks Over Traditional Books
2. Identifying **2 Exploring The Periodic Table Answer Key**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **2 Exploring The Periodic Table Answer Key**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **2 Exploring The Periodic Table Answer Key**
 - Personalized Recommendations
 - **2 Exploring The Periodic Table Answer Key** User Reviews and Ratings

- 2 Exploring The Periodic Table Answer Key and Bestseller Lists
- 5. Accessing 2 Exploring The Periodic Table Answer Key Free and Paid eBooks
 - 2 Exploring The Periodic Table Answer Key Public Domain eBooks
 - 2 Exploring The Periodic Table Answer Key eBook Subscription Services
 - 2 Exploring The Periodic Table Answer Key Budget-Friendly Options
- 6. Navigating 2 Exploring The Periodic Table Answer Key eBook Formats
 - ePUB, PDF, MOBI, and More
 - 2 Exploring The Periodic Table Answer Key Compatibility with Devices
 - 2 Exploring The Periodic Table Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2 Exploring The Periodic Table Answer Key
 - Highlighting and Note-Taking 2 Exploring The Periodic Table Answer Key
 - Interactive Elements 2 Exploring The Periodic Table Answer Key
- 8. Staying Engaged with 2 Exploring The Periodic Table Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2 Exploring The Periodic Table Answer Key
- 9. Balancing eBooks and Physical Books 2 Exploring The Periodic Table Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2 Exploring The Periodic Table Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 2 Exploring The Periodic Table Answer Key
 - Setting Reading Goals 2 Exploring The Periodic Table Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2 Exploring The Periodic Table Answer Key
 - Fact-Checking eBook Content of 2 Exploring The Periodic Table Answer Key
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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