

Lesson Outline**LESSON 2*****Types of Chemical Reactions*****A. Patterns in Reactions**

1. There are _____ major types of chemical reactions.
2. Each type of chemical reaction follows a unique _____ in the way atoms in reactants _____ to form products.

B. Types of Chemical Reactions

1. Knowing the types of chemical reactions helps _____ how substances will react and what _____ will form.
2. In a(n) _____ reaction, two or more substances combine.
 - a. The product in a synthesis reaction is a(n) _____.
 - b. A synthesis reaction has _____ reactant(s) and _____ product(s).
3. In a(n) _____ reaction, a substance breaks down and forms two or more substances.
 - a. The reactant in a decomposition reaction is a(n) _____.
 - b. The products in a decomposition reaction can be _____ or compounds.
 - c. A decomposition reaction has _____ reactant(s) and _____ product(s).
4. In a(n) _____ reaction, an atom or group of atoms replaces part of a(n) _____.
5. One type of replacement reaction is _____ replacement.
 - a. In a(n) _____ reaction, one _____ replaces another element in a compound.
 - b. The reactants in this type of reaction are a(n) _____ and a(n) _____.
 - c. The products in this type of reaction are a different _____ and a different _____.
6. The other type of replacement reaction is _____ replacement.
 - a. In a(n) _____ reaction, the _____ ions in two compounds switch places, forming two new compounds.

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Ralph Thomas Myers

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