

Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)

Name of Reaction Type

1. $2 \text{Na} + \text{Cl}_2 \rightarrow 2 \text{NaCl}$ EX: A B → AB (A & B combine)	Ex: synthesis
2. $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$	
3. $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$	
4. $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$	
5. $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$	
6. $\text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$	
7. $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$	
8. $\text{C}_6\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$	
9. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$	
10. $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2 \text{Ag}$	
11. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$	

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Michael Rajnik

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for Introductory Chemistry by Steve Russo, Mike Silver Saundra Yancy McGuire,Steve Russo, Mike Silver, 2002 Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills **The Pharmaceutical Era** ,1892 **Edexcel International GCSE (9-1) Chemistry Student Book (Edexcel International GCSE (9-1))** Collins,2021-11-12 Exam Board Edexcel Level Subject International GCSE Biology and Double Award Science First teaching September 2017 First exams June 2019 **Scientific American** ,1879

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