

Identifying the 5 Types of Chemical ReactionsIdentify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)Name of Reaction Type

1.	$2 \text{ Na} + \text{Cl}_2 \rightarrow 2 \text{ NaCl}$ EX: A B → AB (A & B combine)	Ex: synthesis
2.	$\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$	
3.	$\text{P}_4 + 3 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_3$	
4.	$2 \text{ NO}_2 \rightarrow 2 \text{ O}_2 + \text{N}_2$	
5.	$\text{Na}_3\text{PO}_4 + 3 \text{ KOH} \rightarrow 3 \text{ NaOH} + \text{K}_3\text{PO}_4$	
6.	$\text{C}_3\text{H}_6\text{O} + 4 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 3 \text{ H}_2\text{O}$	
7.	$\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{ LiCl}$	
8.	$\text{C}_6\text{H}_{12} + 9 \text{ O}_2 \rightarrow 6 \text{ CO}_2 + 6 \text{ H}_2\text{O}$	
9.	$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$	
10.	$2 \text{ AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2 \text{ Ag}$	
11.	$2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$	

1chemical Reactions Practice Problems Answers

Michael Rajnik



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Chemistry, Problem-Solving Worktext James E. Brady, Frederick Senese, 2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes both chemical and physical that they undergo Fireworks One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light Carbon The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C_2H_2 and O_2 to give CO_2 and H_2O Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms Stormy weather The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet

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for Introductory Chemistry by Steve Russo, Mike Silver Sandra Yancy McGuire, Steve Russo, Mike Silver, 2002 Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills

The Pharmaceutical Era, 1892 **Edexcel International GCSE (9-1) Chemistry Student Book (Edexcel International GCSE (9-1))** Collins, 2021-11-12 Exam Board Edexcel Level Subject International GCSE Biology and Double Award Science First teaching September 2017 First exams June 2019 **Scientific American**, 1879

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