

**General Chemistry – Unit 1 Worksheet 6****Dimensional Analysis**

Use the factor-label method to make the following conversions. Remember to use the appropriate number of s.f's in your answer.

**Part 1**

1.  $74 \text{ cm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ meters}$

2.  $8.32 \times 10^{-2} \text{ kg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ grams}$

3.  $55.5 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ cm}^3$

4.  $0.00527 \text{ cal} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ kilocalories}$

5.  $9.52 \times 10^{-4} \text{ m} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ micrometers}$

6.  $41.0 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ liters}$

7.  $6.0 \times 10^{-1} \text{ g} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ mg}$

8.  $8.34 \times 10^{-9} \text{ cg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ g}$

9.  $5.0 \times 10^3 \text{ mm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ m}$

10.  $1 \text{ day} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ seconds}$

11.  $5 \times 10^4 \text{ mm} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ km}$

12.  $9.1 \times 10^{-13} \text{ kg} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ mg}$

13.  $1 \text{ year} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ hours (approximately)}$

# Answer Key Chemistry Unit 6

**Roman Wölfel**

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