

### Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds:

- 1)  $\text{NH}_4\text{Cl}$  \_\_\_\_\_
- 2)  $\text{Fe}(\text{NO}_3)_3$  \_\_\_\_\_
- 3)  $\text{TiBr}_3$  \_\_\_\_\_
- 4)  $\text{Cu}_3\text{P}$  \_\_\_\_\_
- 5)  $\text{SnSe}_2$  \_\_\_\_\_
- 6)  $\text{GaAs}$  \_\_\_\_\_
- 7)  $\text{Pb}(\text{SO}_4)_2$  \_\_\_\_\_
- 8)  $\text{Be}(\text{HCO}_3)_2$  \_\_\_\_\_
- 9)  $\text{Mn}_2(\text{SO}_4)_3$  \_\_\_\_\_
- 10)  $\text{Al}(\text{CN})_3$  \_\_\_\_\_

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate \_\_\_\_\_
- 12) vanadium (IV) carbonate \_\_\_\_\_
- 13) tin (II) nitrite \_\_\_\_\_
- 14) cobalt (III) oxide \_\_\_\_\_
- 15) titanium (II) acetate \_\_\_\_\_
- 16) vanadium (V) sulfide \_\_\_\_\_
- 17) chromium (III) hydroxide \_\_\_\_\_
- 18) lithium iodide \_\_\_\_\_
- 19) lead (II) nitride \_\_\_\_\_
- 20) silver bromide \_\_\_\_\_

### Lots of Ionic Naming Practice Problems

Name the following ionic compounds:

- 1)  $\text{NaBr}$  \_\_\_\_\_
- 2)  $\text{Sr}(\text{OH})_2$  \_\_\_\_\_
- 3)  $\text{V}_2(\text{SO}_4)_3$  \_\_\_\_\_
- 4)  $\text{NH}_4\text{F}$  \_\_\_\_\_
- 5)  $\text{CaCO}_3$  \_\_\_\_\_
- 6)  $\text{NaPO}_4$  \_\_\_\_\_
- 7)  $\text{Li}_2\text{SO}_4$  \_\_\_\_\_
- 8)  $\text{Zn}_2\text{P}_2$  \_\_\_\_\_
- 9)  $\text{Sr}(\text{C}_2\text{H}_5\text{O}_2)_2$  \_\_\_\_\_
- 10)  $\text{Ca}_2\text{O}_4$  \_\_\_\_\_
- 11)  $\text{Ag}_3\text{PO}_4$  \_\_\_\_\_
- 12)  $\text{YCO}_3$  \_\_\_\_\_
- 13)  $\text{SnS}_2$  \_\_\_\_\_
- 14)  $\text{Ti}(\text{CN})_2$  \_\_\_\_\_
- 15)  $\text{KMnO}_4$  \_\_\_\_\_
- 16)  $\text{Pb}_2\text{N}_3$  \_\_\_\_\_
- 17)  $\text{CeCO}_3$  \_\_\_\_\_
- 18)  $\text{CsSO}_4$  \_\_\_\_\_
- 19)  $\text{Cu}(\text{NO}_3)_2$  \_\_\_\_\_
- 20)  $\text{Fe}(\text{HCO}_3)_2$  \_\_\_\_\_

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