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Synthesis & Decomposition Reactions

40 points

Part 1

Name _____

Date _____

- A. Below the word equation write the chemical equation for the reaction.
- B. Determine if the reaction is a Synthesis or Decomposition reaction.
- C. Copper reacts with Chlorine gas to produce Copper (II) chloride.



• Synthesis

- A. Ammonium Chloride reacts with Hydrochloric acid (HCl) to produce Ammonium chloride.



• Synthesis

- A. Aluminum Hydroxide produces Aluminum oxide and water.



• Decomposition

- A. Sodium metal reacts with Oxygen gas to produce Sodium oxide.



• Synthesis

- A. Copper (II) oxide produces Copper metal and Oxygen gas.



Answer Key To Decomposition Synthesis Reactions

Raffaela Di Napoli

Answer Key To Decomposition Synthesis Reactions:

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