

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON-METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H₂ hydrogen is diatomic:2.) F₂ fluorine is diatomic:3.) O₂ oxygen is diatomic:4.) N₂ nitrogen is diatomic. Is this a triple bond?5.) BF₃ you need 3 fluorine atoms here:6.) Ammonia NH₃ hint: how many hydrogen atoms are needed?7.) Carbon dioxide CO₂8.) Methane CH₄ careful here 4 hydrogen atoms needed

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

10.) SO₂ hint: one pair of electrons from sulfur must be split up for this one to work.

8 2 Covalent Bonding Worksheet Answers

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