

38. α -amino acids, $\text{R}-\underset{\text{NH}_2}{\text{CH}}-\text{COOH}$

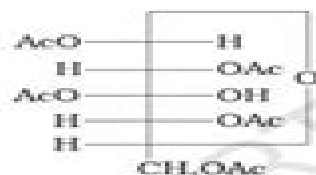
39. In α -helix, a polypeptide chain is stabilised by the formation of hydrogen bonds between $-\text{NH}-$ group of amino acids in one turn with the $>\text{C}=\text{O}$ groups of amino acids belonging to adjacent turn.

40. Oxidoreductase

41. Lactic acid.

42. Glucose gives pentaacetate derivative on acetylation with acetic anhydride. This confirms the presence of five $-\text{OH}$ groups.

43. Glucose pentaacetate (structure A) doesn't have a free $-\text{OH}$ group at C1 and so can't be converted to the open chain form to give $-\text{CHO}$ group and hence doesn't form the osime.



(Structure A)

44. Vitamin C is water soluble therefore it is readily excreted in urine and can't be stored in our body.

45. On hydrolysis sucrose (dextrorotatory), gives glucose (dextrorotatory, $+52.5^\circ$) and fructose (laevorotatory, -92.4°). Since laevorotation of fructose is more than the dextrorotation of glucose, the mixture is laevorotatory.

46. In aqueous solution, the carboxyl group loses a proton and amino group accepts a proton to form a zwitter ion.



47. In glycylalanine, carboxyl group of glycine combines with the amino group of alanine.



14 Exemplar Chemistry Grd1

Guillaume Favre



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