

Multiple Choice

Select the alternative that correctly completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$, what is the molar enthalpy change of expansion?
- ☒ 100
☐ 200
☐ 300
☐ 400

3. The enthalpy of fusion is the lowest for:
- ☐ H₂O
☐ H₂O (ice)
☒ H₂O (liquid)
☒ H₂O (solid)

4. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the heat of reaction is 285 kJ/mol. The heat of reaction is 285 kJ/mol. The heat of reaction is 285 kJ/mol.
- ☐ 285 kJ/mol
☐ 570 kJ/mol
☒ 1125 kJ/mol
☐ 2250 kJ/mol

5. What is the value of the standard enthalpy of formation of water?
- ☐ 285 kJ/mol
☒ 1125 kJ/mol
☐ 2250 kJ/mol
☐ 4500 kJ/mol

6. What is the heat of fusion of water?
- ☐ 285 kJ/mol
☐ 570 kJ/mol
☒ 1125 kJ/mol
☐ 2250 kJ/mol

7. A sample of water is heated from 0°C to 100°C. The heat of fusion is 333 J/g. The heat of fusion is 333 J/g.
- ☒ 333 J/g
☐ 666 J/g
☐ 999 J/g
☐ 1332 J/g

8. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the heat of reaction is 285 kJ/mol. The heat of reaction is 285 kJ/mol.
- ☒ 285 kJ/mol
☐ 570 kJ/mol
☐ 1125 kJ/mol
☐ 2250 kJ/mol

9. At constant $T_1 = 100^\circ\text{C}$, what is the molar enthalpy change of expansion?
- ☐ 100
☐ 200
☒ 300
☐ 400

10. The enthalpy of fusion is the lowest for:
- ☐ H₂O
☒ H₂O (ice)
☐ H₂O (liquid)
☐ H₂O (solid)

11. The enthalpy of fusion is the lowest for:
- ☐ H₂O
☒ H₂O (ice)
☐ H₂O (liquid)
☐ H₂O (solid)

Multiple Choice Stoichiometry Test

Björn Risch



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countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems

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