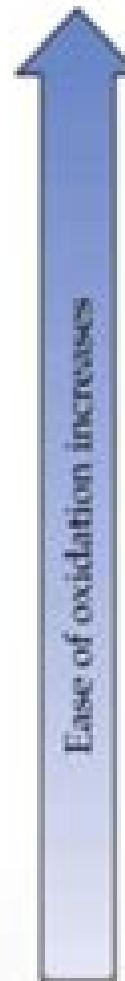


The Activity Series. Topic 9.1

Metal	Oxidation Reaction
Lithium	$\text{Li(s)} \longrightarrow \text{Li}^+(aq) + \text{e}^-$
Potassium	$\text{K(s)} \longrightarrow \text{K}^+(aq) + \text{e}^-$
Barium	$\text{Ba(s)} \longrightarrow \text{Ba}^{2+}(aq) + 2\text{e}^-$
Calcium	$\text{Ca(s)} \longrightarrow \text{Ca}^{2+}(aq) + 2\text{e}^-$
Sodium	$\text{Na(s)} \longrightarrow \text{Na}^+(aq) + \text{e}^-$
Magnesium	$\text{Mg(s)} \longrightarrow \text{Mg}^{2+}(aq) + 2\text{e}^-$
Aluminum	$\text{Al(s)} \longrightarrow \text{Al}^{3+}(aq) + 3\text{e}^-$
Manganese	$\text{Mn(s)} \longrightarrow \text{Mn}^{2+}(aq) + 2\text{e}^-$
Zinc	$\text{Zn(s)} \longrightarrow \text{Zn}^{2+}(aq) + 2\text{e}^-$
Chromium	$\text{Cr(s)} \longrightarrow \text{Cr}^{3+}(aq) + 3\text{e}^-$
Iron	$\text{Fe(s)} \longrightarrow \text{Fe}^{2+}(aq) + 2\text{e}^-$
Cobalt	$\text{Co(s)} \longrightarrow \text{Co}^{2+}(aq) + 2\text{e}^-$
Nickel	$\text{Ni(s)} \longrightarrow \text{Ni}^{2+}(aq) + 2\text{e}^-$
Tin	$\text{Sn(s)} \longrightarrow \text{Sn}^{2+}(aq) + 2\text{e}^-$
Lead	$\text{Pb(s)} \longrightarrow \text{Pb}^{2+}(aq) + 2\text{e}^-$
Hydrogen	$\text{H}_2\text{(g)} \longrightarrow 2\text{H}^+(aq) + 2\text{e}^-$
Copper	$\text{Cu(s)} \longrightarrow \text{Cu}^{2+}(aq) + 2\text{e}^-$
Silver	$\text{Ag(s)} \longrightarrow \text{Ag}^+(aq) + \text{e}^-$
Mercury	$\text{Hg(l)} \longrightarrow \text{Hg}^{2+}(aq) + 2\text{e}^-$
Platinum	$\text{Pt(s)} \longrightarrow \text{Pt}^{2+}(aq) + 2\text{e}^-$
Gold	$\text{Au(s)} \longrightarrow \text{Au}^{3+}(aq) + 3\text{e}^-$



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