

Relative formula mass

Practice worksheet

4.1 Calculating relative formula mass (Mr)	
Calculate the relative molecular mass of water $M_r(H_2O) = 2 \times 1 + 16 = 18$ $= 2 \times 1 + 2 \times 16$ $= 2 + 32$ $= 34$	

Calculate the Mr of MgO

$$\begin{aligned}M_r(MgO) &= 1 \times Mg + 1 \times O \\&= 1 \times 24 + 1 \times 16 \\&= 24 + 16 \\&= 40\end{aligned}$$

Calculate the Mr of C₂H₆

$$\begin{aligned}M_r(C_2H_6) &= 2 \times C + 6 \times H \\&= 2 \times 12 + 6 \times 1 \\&= 24 + 6 \\&= 30\end{aligned}$$

Calculate the Mr of H₂SO₄

$$\begin{aligned}M_r(H_2SO_4) &= 2 \times H + 1 \times S + 4 \times O \\&= 2 \times 1 + 1 \times 32 + 4 \times 16 \\&= 2 + 32 + 64 \\&= 98\end{aligned}$$

Calculate the Mr of CH₄

$$\begin{aligned}M_r(CH_4) &= 1 \times C + 4 \times H \\&= 1 \times 12 + 4 \times 1 \\&= 12 + 4 \\&= 16\end{aligned}$$

Calculate the Mr of CaCO₃

$$\begin{aligned}M_r(CaCO_3) &= 1 \times Ca + 1 \times C + 3 \times O \\&= 1 \times 40 + 1 \times 12 + 3 \times 16 \\&= 40 + 12 + 48 \\&= 92\end{aligned}$$

Calculate the Mr of Al₂O₃

$$\begin{aligned}M_r(Al_2O_3) &= 2 \times Al + 3 \times O \\&= 2 \times 27 + 3 \times 16 \\&= 54 + 48 \\&= 102\end{aligned}$$

Calculate the Mr of K₂SO₄

$$\begin{aligned}M_r(K_2SO_4) &= 2 \times K + 1 \times S + 4 \times O \\&= 2 \times 39 + 1 \times 32 + 4 \times 16 \\&= 78 + 32 + 64 \\&= 174\end{aligned}$$

4.1 Calculating relative formula mass (Mr)	
Calculate the Mr of Al ₂ O ₃ $M_r(Al_2O_3) = 2 \times Al + 3 \times O$ $= 2 \times 27 + 3 \times 16$ $= 54 + 48$ $= 102$	Calculate the Mr of CaCO ₃ $M_r(CaCO_3) = 1 \times Ca + 1 \times C + 3 \times O$ $= 1 \times 40 + 1 \times 12 + 3 \times 16$ $= 40 + 12 + 48$ $= 92$
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Activity Relative Mass Answers

RJ Shavelson

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