

Atomic Structure



Atomic Number

Element Symbol

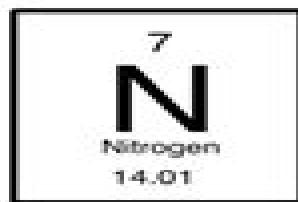
Element Name

Atomic Mass

The atomic number represents these two parts of the atom:
number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

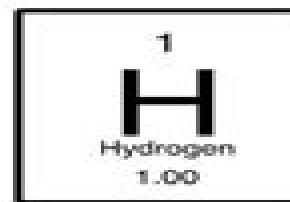
7

Electrons:

7

Neutrons:

7



Protons:

1

Electrons:

1

Neutrons:

0



Protons:

8

Electrons:

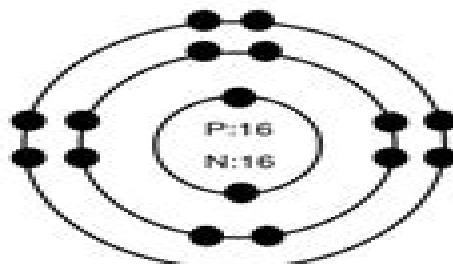
8

Neutrons:

8

BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

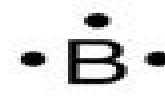
Element: SulfurNumber of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



Atomic Structure Review Worksheet Answer Key

Jo Locke

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Table of Contents Atomic Structure Review Worksheet Answer Key

1. Understanding the eBook Atomic Structure Review Worksheet Answer Key
 - The Rise of Digital Reading Atomic Structure Review Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Review Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Review Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Review Worksheet Answer Key
 - Personalized Recommendations
 - Atomic Structure Review Worksheet Answer Key User Reviews and Ratings
 - Atomic Structure Review Worksheet Answer Key and Bestseller Lists
5. Accessing Atomic Structure Review Worksheet Answer Key Free and Paid eBooks

- Atomic Structure Review Worksheet Answer Key Public Domain eBooks
- Atomic Structure Review Worksheet Answer Key eBook Subscription Services
- Atomic Structure Review Worksheet Answer Key Budget-Friendly Options

6. Navigating Atomic Structure Review Worksheet Answer Key eBook Formats

- ePUB, PDF, MOBI, and More
- Atomic Structure Review Worksheet Answer Key Compatibility with Devices
- Atomic Structure Review Worksheet Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Atomic Structure Review Worksheet Answer Key
- Highlighting and Note-Taking Atomic Structure Review Worksheet Answer Key
- Interactive Elements Atomic Structure Review Worksheet Answer Key

8. Staying Engaged with Atomic Structure Review Worksheet Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Atomic Structure Review Worksheet Answer Key

9. Balancing eBooks and Physical Books Atomic Structure Review Worksheet Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Atomic Structure Review Worksheet Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Atomic Structure Review Worksheet Answer Key

- Setting Reading Goals Atomic Structure Review Worksheet Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Structure Review Worksheet Answer Key

- Fact-Checking eBook Content of Atomic Structure Review Worksheet Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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