

## Experiment 11

## **Data and Calculations: The Atomic Spectrum of Hydrogen**

#### A. Calculation of the Energy Levels of the Hydrogen Atom

Reserves are also calculated from Equations 6 and 7 for the 10 lowest energy stations.

Table 11.2

Quantum number, $n$	Energy, $E$ , (kJ/mol)	Quantum number, $n$	Energy, $E$ , (kJ/mol)

## B. Calculation of the Wavelengths of the Lines in the Hydrogen Atomic Spectrum

In the upper half of each box, write  $\Delta E$ , the difference in energy in kJ/mole between  $E_{\infty}$  and  $E_{\text{rec}}$ . In the lower half of the box, write the  $\Delta$  in nm associated with that value of  $\Delta E$ .

Table 11.3

	6	5	4	3	2	1
1	12.75. 16 KJ/mol	1250.56 KJ/mol	12.30. 04 KJ/mol	11.66. 25 KJ/mol	9.74. 05 KJ/mol	
	93.78 nm	94.91 nm	91.15 nm	101.51 nm	121.56 nm	
2	2.91. 54 KJ/mol	235.53 KJ/mol	2.46. 01 KJ/mol	1.82. 23 KJ/mol		
	410.23 nm	434.17 nm	421.26 nm	456.46 nm		
3	109.34 KJ/mol	93.3 KJ/mol	63.73 KJ/mol			
	1694.63 nm	1251.11 nm	1315.61 nm			
4	4.56 KJ/mol	29.52 KJ/mol				
	1675.10 nm	4052.46 nm				
5	16.04 KJ/mol					
	7452.64 nm					

$$\Delta E = E_{\text{final}} - E_{\text{initial}}$$

$$\lambda \text{ [nm]} = \frac{1.19627 \times 10^9}{\Delta E}$$

(continued on following page)

# Atomic Spectrum Of Hydrogen Lab Answer Key

**K Morrison**

**Atomic Spectrum Of Hydrogen Lab Answer Key:**

The book delves into Atomic Spectrum Of Hydrogen Lab Answer Key. Atomic Spectrum Of Hydrogen Lab Answer Key is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Atomic Spectrum Of Hydrogen Lab Answer Key, encompassing both the fundamentals and more intricate discussions.

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  - Chapter 1: Introduction to Atomic Spectrum Of Hydrogen Lab Answer Key
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  - Chapter 3: Atomic Spectrum Of Hydrogen Lab Answer Key in Everyday Life
  - Chapter 4: Atomic Spectrum Of Hydrogen Lab Answer Key in Specific Contexts
  - Chapter 5: Conclusion
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