

Chapter 4

Atomic Structure

Class 8 - Dalal Simplified Middle School Chemistry Solutions

The atom - ancient views

Question 1

State the main postulates of - Dalton's atomic theory. Explain how the modern atomic theory contradicted Dalton's atomic theory.

Answer

Postulates of Dalton's atomic theory : Matter consists of particles called - atoms, which are indivisible & cannot be created or destroyed.

1. Atoms of the same element are alike but differ from atoms of all other elements.
2. Compounds or molecules are formed when atoms combine with other atoms in simple whole number ratios.
3. An atom is the smallest unit of matter which takes part in a chemical reaction.

Contradiction :

Modern atomic theory
Atoms are destructible
Atoms are divisible & consist of - protons, electrons & neutrons
Atoms of the same element may not be alike as seen in case of isotopes

Dalton's atomic theory
Atoms are indestructible
Atoms are indivisible & smallest particle of an element.
Atoms of the same element are alike in all respect.

Sub-atomic particles

Question 2

With reference to the discovery of the structure of an atom, explain in brief - William Crookes experiment for the discovery of cathode rays, followed by - J.J. Thomson's

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H Kauffman



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