

7. Atomic Spectra Experiment

Preparation

In this experiment, you will build a FPGAs spectrometer to analyze the emission spectrum of a hydrogen discharge lamp (HDL).

Theory

Atomic spectra measurements and analysis strategy relies on a semi-empirical analysis using various scientific methods such as quantum electrodynamics (QED), perturbation theory, and statistical methods. The analysis of the spectrum of a hydrogen discharge lamp is a complex task. The analysis of the spectrum of a hydrogen discharge lamp is a complex task. The analysis of the spectrum of a hydrogen discharge lamp is a complex task. The analysis of the spectrum of a hydrogen discharge lamp is a complex task.

Using spectral spectroscopy measurements, the energy of the spectrum emitted by the atom can be determined.

$$E_{\text{photon}} = \frac{hc}{\lambda} \quad (1)$$

where h is the Planck constant, c is the speed of light, and λ is the wavelength of the electromagnetic wave.

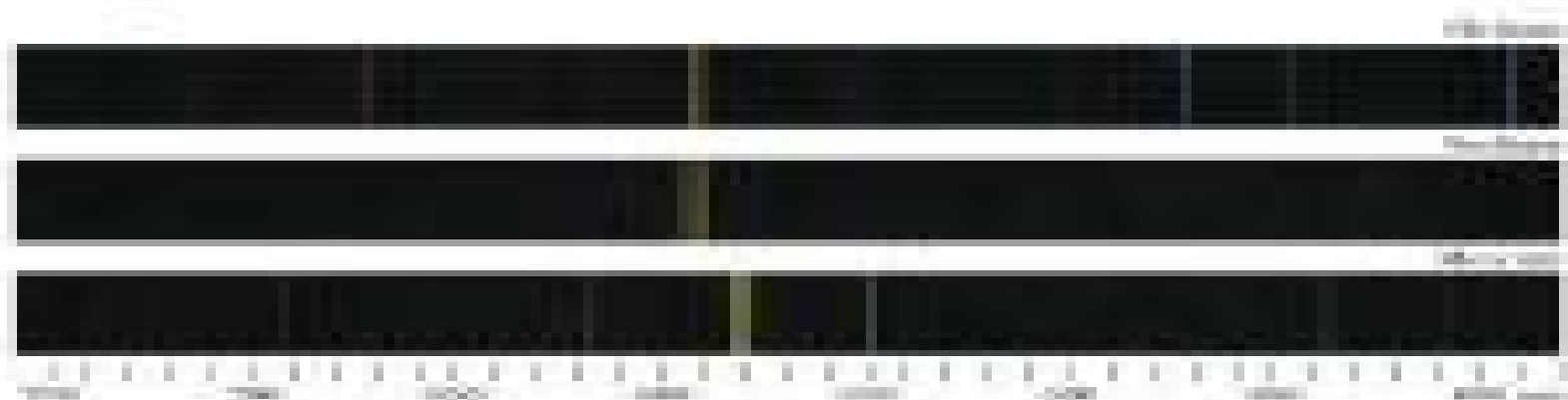


Figure 1. Emission spectrum of the HDL (hydrogen discharge lamp). The spectrum is shown with the wavelength (nm) on the x-axis and the intensity (arbitrary units) on the y-axis. The spectrum shows a prominent yellow line at approximately 656 nm, which is the characteristic emission of the hydrogen atom. The spectrum also shows other lines, including a red line at approximately 656 nm and a blue line at approximately 486 nm.

Atomic Spectra Lab Answers

Jared M. Diamond, Ted J. Case



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