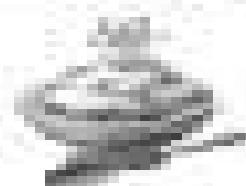


Atomic Probing Scales of the Atom

Name _____
Date _____



Purpose:

To determine which chemical elements are represented by the gas and liquid phases.

Part I: Using Scales to Determine Phases

When a substance is placed in a container, it will either be a solid, liquid, or gas. The state of a substance is determined by its temperature and pressure. In this experiment, you will use a scale to determine the state of a substance by measuring its weight and volume.

Procedure: 1. Weigh a small amount of the substance (about 1 gram) and record the weight.

2. Place the substance in a container and measure its volume. Record the volume.

Results: The substance is a solid because it has a definite shape and volume. The substance is a liquid because it has a definite volume but no definite shape. The substance is a gas because it has no definite shape or volume.

3. Repeat the experiment for the other substances. Record the results.

Conclusion: The substance is a solid because it has a definite shape and volume. The substance is a liquid because it has a definite volume but no definite shape. The substance is a gas because it has no definite shape or volume.

4. Repeat the experiment for the other substances. Record the results.

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