

# Atomic Mass Atomic Number Worksheet

Name: \_\_\_\_\_

Date: \_\_\_\_\_

Name of Element	Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
copper	Cu	29	64	29	35	29
tin	Sn	50	119	50	69	50
iodine	I	53	127	53	74	53
uranium	U	92	238	92	146	92
potassium	K	19	39	19	20	19
lithium	Li	3	7	3	4	3
oxygen	O	8	16	8	8	8
gold	Au	79	197	79	118	79
sulfur	S	16	32	16	16	16
silver	Ag	47	108	47	61	47
chromium	Cr	24	52	24	28	24
cobalt	Co	27	59	27	32	27
nickel	Ni	28	59	28	31	28
zinc	Zn	30	65	30	35	30
aluminum	Al	13	27	13	14	13
mercury	Hg	80	201	80	121	80
platinum	Pt	78	195	78	117	78
iron	Fe	26	56	26	30	26
hydrogen	H	1	1	1	0	1
helium	He	2	4	2	2	2
beryllium	Be	4	9	4	5	4
magnesium	Mg	12	24	12	12	12
carbon	C	6	12	6	6	6
silicon	Si	14	28	14	14	14
chlorine	Cl	17	35	17	18	17
bismuth	Bi	83	209	83	126	83
boron	B	5	11	5	6	5
calcium	Ca	20	40	20	20	20
manganese	Mn	25	55	25	30	25
lead	Pb	82	207	82	125	82
sodium	Na	11	23	11	12	11
fluorine	F	9	19	9	10	9
phosphorus	P	15	31	15	16	15

# Atomic Mass Atomic Number Answer Key

**Stephen Glazier**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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