

CLASS:XI CHEMISTRY
CHAPTER#02
ATOMIC STRUCTURE
BY: SIR AAMIR HULLIO

1. Cathode rays are deflected by
(a) electric field only (b) magnetic field only (c) electric and magnetic field (d) none of these
2. The e/m value for the particles constituting cathode rays is the same regardless of
(a) the gas present in cathode rays tube (b) the metal of which cathode was made
(c) both of these (d) none of these
3. The charge to mass ratio (e/m) of positive particles
(a) varies with the nature of gas in discharge tube (b) is independent of the gas in discharge tube
(c) is constant (d) none of the above
4. A sub atomic particle which has one unit mass and one unit positive charge is known as
(a) hydrogen atom (b) neutron (c) electron (d) proton
5. Atomic number of an element is equal to the number of _____ in the nucleus of the atom.
(a) neutrons (b) protons (c) both the neutrons and protons (d) electrons
6. The mass number of an atom is equal to the number of _____ in the nucleus of an atom
(a) protons (b) neutrons (c) electrons (d) nucleons
7. The mass number and atomic number of Phosphorus atom are 31 and 15 respectively. The number of neutrons in the nucleus is
(a) 15 (b) 16 (c) 31 (d) 46
8. In a sodium atom (atomic number = 11 and mass number = 23), the number of neutrons is
(a) equal to the number of protons (b) less than the number of protons
(c) greater than the number of protons (d) none of these
9. Which of the following is not correct for electromagnetic waves?
(a) the wavelength is the distance between two successive crests
(b) the frequency is the number of waves which pass a given point in one second
(c) the velocity of a wave is the distance covered by the particular wave in one second
(d) all electromagnetic waves have equal wavelengths
10. If Z is the number of proton and A the number of nucleons, then the number of neutrons in an atom is given by
(a) $A + Z$ (b) $A - Z$ (c) $Z - A$ (d) none of these
11. The mass number and atomic number of Phosphorus atom are 31 and 15 respectively. The number of neutrons in the nucleus is
(a) 15 (b) 16 (c) 31 (d) 46
12. Which of the following relations is not correct?
(a) $c = \lambda \times \mu$ (b) $\mu = 1/f$ (c) $\mu = 1/\lambda$ (d) $\lambda = c + \mu$
13. The unit in which wave number is measured
(a) hertz (b) sec^{-1} (c) nanometer (d) cm^{-1}
14. In the spectrum of hydrogen atom, the series which falls in ultraviolet region is
(a) Lyman series (b) Balmer series (c) Paschen series (d) Brackett series
15. The Balmer series in the spectrum of hydrogen atom falls in
(a) ultraviolet region (b) visible region (c) infrared region (d) none of these
16. The radius of first orbit in hydrogen atom is 0.529 Å. The radius of second orbit is given by
(a) $\frac{1}{2} \times 0.529 \text{ \AA}$ (b) $2 \times 0.529 \text{ \AA}$ (c) $4 \times 0.529 \text{ \AA}$ (d) $8 \times 0.529 \text{ \AA}$

Atomic Structure Ch 4 Packet Answers

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**

Atomic Structure Ch 4 Packet Answers:

Organic Chemistry, 13e Student Study Guide and Solutions Manual T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text [Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual](#) T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us **Organic Chemistry, 5e Student Study Guide and Solutions Manual** David R. Klein,Laurie S. Starkey,2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry [Organic Chemistry, 4e Student Solution Manual and Study Guide](#) David R. Klein,2021-01-07 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Solution Manual and Study Guide 4th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

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Study Guide, Chemical Principles, Fifth Edition, Zumdahl Paul B. Kelter, Steven S. Zumdahl, 2004-04

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