

Name _____ Group _____

Chemistry 2nd Semester
Review 4th Period exam

1.- Identify as organic or inorganic

- a) KCl b) C₂H₆O c) CaCl₂ d) CO₂ e) NaCN f) CH₄

2.- Identify the following properties as organic or inorganic

- a) soluble in water b) burns in air c) low boiling point d) contain K
e) form ionic bond f) form isomers g) high boiling point h) can be used as fuel

3.- True or false

- a) Carbon atoms can bond with other carbon atoms any number of times
b) Carbon atomic number is 4
c) Carbon forms 4 covalent bonds
d) Carbon can bond with many elements, metals and nonmetals

4.- Identify the following as alkane, alkene or alkyne

- a) $\begin{array}{c} \text{CH}_3\text{CH}_3 \\ | \\ \text{HC}\equiv\text{CHCH}_2\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$ b) $\begin{array}{c} \text{CH}_3-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$ c) CH₂=CH-CH₃

5.- Identify as saturated or unsaturated

- a) $\begin{array}{c} \text{CH}_3-\text{CH}-\text{C}=\text{CH}_2 \\ | \quad | \\ \text{CH}_3 \quad \text{CH}_3 \end{array}$ b) $\begin{array}{c} \text{H}_2\text{C}-\text{CH}-\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$

6.- Give the name of the following alkyl groups

- a) -CH₃ b) -CH₂-CH₃ c) -CH₂-CH₂-CH₃ d) -CH₃-CH-CH₃

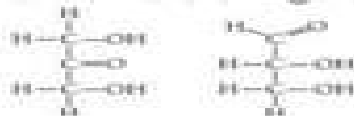
7.- What are aliphatic hydrocarbons.

8.- Identify as hydrocarbon or derivate of hydrocarbon

- a) CH₃-Cl b) CH₃-CH₂-CH₃ c) CH₃-COOH d) CH₃-CH=CH-CH₃
e) CH₃-O-CH₃ f) C₆H₆ g) CH₃-CH=O

9.- What are isomers?

10.- Do the following molecules represent isomers? Explain.



11.- Circle the functional group and identify the class or family.

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**Giles Murrel Ruch, George Dinsmore
Stoddard**



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Table of Contents Nd Semester Exam Chemistry Form

1. Understanding the eBook Nd Semester Exam Chemistry Form
 - The Rise of Digital Reading Nd Semester Exam Chemistry Form
 - Advantages of eBooks Over Traditional Books
2. Identifying Nd Semester Exam Chemistry Form
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nd Semester Exam Chemistry Form
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nd Semester Exam Chemistry Form
 - Personalized Recommendations

- Nd Semester Exam Chemistry Form User Reviews and Ratings
- Nd Semester Exam Chemistry Form and Bestseller Lists
- 5. Accessing Nd Semester Exam Chemistry Form Free and Paid eBooks
 - Nd Semester Exam Chemistry Form Public Domain eBooks
 - Nd Semester Exam Chemistry Form eBook Subscription Services
 - Nd Semester Exam Chemistry Form Budget-Friendly Options
- 6. Navigating Nd Semester Exam Chemistry Form eBook Formats
 - ePub, PDF, MOBI, and More
 - Nd Semester Exam Chemistry Form Compatibility with Devices
 - Nd Semester Exam Chemistry Form Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nd Semester Exam Chemistry Form
 - Highlighting and Note-Taking Nd Semester Exam Chemistry Form
 - Interactive Elements Nd Semester Exam Chemistry Form
- 8. Staying Engaged with Nd Semester Exam Chemistry Form
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nd Semester Exam Chemistry Form
- 9. Balancing eBooks and Physical Books Nd Semester Exam Chemistry Form
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nd Semester Exam Chemistry Form
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nd Semester Exam Chemistry Form
 - Setting Reading Goals Nd Semester Exam Chemistry Form
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nd Semester Exam Chemistry Form
 - Fact-Checking eBook Content of Nd Semester Exam Chemistry Form

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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