

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 200°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$, what is the molar volume of hydrogen gas?
- ☒ 20
☐ 10
☐ 5
☐ 1

3. The molar volume of hydrogen is the same as:
- ☐ that of any gas
☐ that of any liquid
☒ that of any solid
☒ that of any gas

4. For hydrogen, $T_1 = 100^\circ\text{C}$, $T_2 = 200^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 200$ atm, what is the molar volume of hydrogen gas?
- ☐ 10
☐ 20
☐ 5
☒ 1

5. What is the molar volume of hydrogen gas at 100 atm, 200°C?
- ☐ 10
☒ 20
☐ 5
☐ 1

6. What is the molar volume of hydrogen gas at 100 atm, 200°C?
- ☐ 10
☐ 20
☐ 5
☒ 1

7. A cylinder of hydrogen gas at 100 atm, 200°C, is used to produce hydrogen gas. What is the molar volume of hydrogen gas?
- ☒ 10
☐ 20
☐ 5
☐ 1

8. For hydrogen, $T_1 = 100^\circ\text{C}$, $T_2 = 200^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 200$ atm, what is the molar volume of hydrogen gas?
- ☒ 10
☐ 20
☐ 5
☐ 1

9. At constant $T_1 = 100^\circ\text{C}$, $T_2 = 200^\circ\text{C}$, $P_1 = 100$ atm, $P_2 = 200$ atm, what is the molar volume of hydrogen gas?
- ☐ 10
☐ 20
☐ 5
☒ 1

10. What is the molar volume of hydrogen gas at 100 atm, 200°C?
- ☐ 10
☒ 20
☐ 5
☐ 1

11. The molar volume of hydrogen gas at 100 atm, 200°C, is the same as the molar volume of any gas.

Multiple Choice Stoichiometry Test

S Marginson



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