

**Review for the Full Year General Chemistry
American Chemical Society Exam**

(a) Thermodynamics, (b) Transition Metals, (c) Redox and Electrochemistry, (d) Nuclear and (e) Organic

THERMODYNAMICS

84. A 10.0 g sample of silver is heated to 100.0 °C and then added to 20.0 g of water at 23.0 °C in an insulated calorimeter. At thermal equilibrium the temperature of the system was measured as 25.0 °C. What is the specific heat of silver? [specific heat = $4.2 \text{ J g}^{-1} \text{ }^{\circ}\text{C}^{-1}$]

- (A) 0.11 J/g K (C) 17 J/g K
(B) 0.22 J/g K (D) 34 J/g K

85. What is ΔH in kJ for the reaction:



- (A) -1830 kJ (B) -2284 kJ
(C) -2510 kJ (D) -1605 kJ

86. Given these equations:



calculate the heat of formation of $\text{SO}_3(\text{g})$.

- (A) -500 kJ mol⁻¹ (C) +100 kJ mol⁻¹
(B) -400 kJ mol⁻¹ (D) +200 kJ mol⁻¹

87. Given:



How much energy is produced in the combustion of 28 g of CO? [Atomic Masses: C 12.0 g mol⁻¹; O 16.0 g mol⁻¹]

- (A) 88 kJ (B) 109 kJ
(C) 175 kJ (D) 284 kJ

88. Calculate the value of ΔH (in kJ mol⁻¹) for the reaction:



Bond Energies (kJ mol ⁻¹)	
H-H	435
N=N	945 (in N ₂)
N-H	389

- (A) 2340 kJ of heat absorbed
(B) 213 kJ of heat absorbed
(C) 2340 kJ of heat evolved
(D) 83 kJ of heat evolved

89. Which change is likely to be accompanied by the greatest increase in entropy?



90. For which process is the entropy change per mole the largest at constant temperature?



91. In which process is entropy decreased?

- (A) dissolving sugar in water
(B) expanding a gas
(C) evaporating a liquid
(D) freezing water

92. When $\text{Al}_2\text{O}_3(\text{s})$ is formed from the elements at standard conditions, the values of ΔH° and ΔG° at 298 K are -1676 kJ mol⁻¹ and -1577 kJ mol⁻¹, respectively. The standard entropy of formation per mole, in joules per degree, will be

- (A) -332 (B) -157 (C) -93.3
(D) -0.0933 (E) +15.7

93. Vaporization of a liquid is an example of a process for which

- (A) ΔH , ΔS , and ΔG are positive at all temperatures.
(B) ΔH and ΔS are positive.
(C) ΔG is negative at low temperatures, positive at high temperatures.
(D) $\Delta H = \Delta S$

94. A particular chemical reaction has a negative ΔH and negative ΔS . Which statement is correct?

- (A) The reaction is spontaneous at all temperatures.
(B) The reaction is nonspontaneous at all temperatures.
(C) The reaction becomes spontaneous as temperature increases.
(D) The reaction becomes spontaneous as temperature decreases.

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Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS, 2018-10-15 This guide is separated into first term and second term general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems **Building**

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Resources in Education, 1995 **Mechanical Engineering**, 1964 **Books in Series, 1876-1949** R.R. Bowker

Company,1982 Preparing for Your ACS Examination in Physical Chemistry Thomas A. Holme,Kristen Murphy,2009
American Druggists' Circular and Chemical Gazette ,1883 Books and Pamphlets, Including Serials and Contributions to Periodicals Library of Congress. Copyright Office,1970 **Guide to Science Teaching in Secondary Schools** Archie L. Lacey,Virginia M. Rowley,1966 **Subject Guide to Books in Print** ,2001 Official Guide to Catholic Educational Institutions and Religious Communities in the United States ,1962 Acs Organic Chemistry Study Guide 2025-2026 - 2 Full-Length Practice Tests, Acs Secrets Exam Prep Book Matthew Bowling,2025-08-09 Mometrix Test Preparation s ACS Organic Chemistry Study Guide ACS Secrets Exam Prep Book is the ideal prep solution for anyone who wants to pass their ACS Organic Chemistry Exam The exam is extremely challenging and thorough test preparation is essential for success Our study guide includes 2 practice tests available in online interactive format All 2 of these printed in the guide Tips and strategies to help you get your best test performance A complete review of all organic chemistry test sections ACS is a registered trademark of the American Chemical Society which is not affiliated with Mometrix Test Preparation and does not endorse this product The Mometrix guide is filled with the critical information you will need in order to do well on your organic chemistry exam the concepts procedures principles and vocabulary that the American Chemical Society ACS Examinations Institute expects you to have mastered before sitting for your exam Sections include Structure Acids and Bases Nucleophilic Substitution Reactions Elimination Reactions Addition and Other Reactions Spectroscopy Radical Reactions Conjugated Systems and Aromaticity Aromatic Reactions Carbonyl Chemistry Enol and Enolate Chemistry Applications and much more Our guide is full of specific and detailed information that will be key to passing your exam Concepts and principles aren t simply named or described in passing but are explained in detail The Mometrix organic chemistry study guide is laid out in a logical and organized fashion so that one section naturally flows from the one preceding it Because it s written with an eye for both technical accuracy and accessibility you will not have to worry about getting lost in dense academic language Any test prep guide is only as good as its practice questions and answer explanations and that s another area where our guide stands out The Mometrix test prep team has provided plenty of organic chemistry practice test questions to prepare you for what to expect on the actual exam Each answer is explained in depth in order to make the principles and reasoning behind it crystal clear All 2 practice tests are available to take in online interactive format allowing you to immediately score your test and see what you got wrong We ve also printed all 2 practice tests in your guide for offline reference We ve helped hundreds of thousands of people pass standardized tests and achieve their education and career goals We ve done this by setting high standards for Mometrix Test Preparation guides and our ACS Organic Chemistry Study Guide ACS Secrets Exam Prep Book is no exception It s an excellent investment in your future Get the organic chemistry review you need to be successful on your exam Iron Age ,1916 **Technical Book Review** ,1965

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