

# ASSIGNING OXIDATION NUMBERS

Name \_\_\_\_\_

Assign oxidation numbers to all of the elements in each of the compounds or ions below.

1. $\text{HCl}$	11. $\text{H}_2\text{SO}_3$
2. $\text{KNO}_3$	12. $\text{H}_2\text{SO}_4$
3. $\text{OH}^-$	13. $\text{BaO}_2$
4. $\text{Mg}_3\text{N}_2$	14. $\text{KMnO}_4$
5. $\text{KClO}_3$	15. $\text{LiH}$
6. $\text{Al}(\text{NO}_3)_3$	16. $\text{MnO}_2$
7. $\text{S}_8$	17. $\text{OF}_2$
8. $\text{H}_2\text{O}_2$	18. $\text{SO}_3$
9. $\text{PbO}_2$	19. $\text{NH}_3$
10. $\text{NaHSO}_4$	20. $\text{Na}$

# Assigning Oxidation Numbers Worksheet Answer Key

**Norman Joseph LaFave**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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