

Here are some practice problems for the current material dealing with molar masses. As an extra challenge, could anyone identify the substances below?

Calculate the formula masses for the compounds below.

a)	H_3PO_4	b)	C_4H_{10}	c)	$\text{Co}(\text{NO}_3)_2$
	98 amu		58 amu		183 amu
d)	FeCl_3	e)	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$	f)	$\text{Al}_2(\text{SO}_4)_3$
	161 amu		342 amu		342 amu

Calculate molar masses for the compounds below.

a)	H_2O	b)	H_2SO_4	c)	$\text{C}_6\text{H}_{12}\text{O}_6$
	18 g/mol		98 g/mol		180 g/mol
d)	$\text{C}_7\text{H}_8\text{N}_4\text{O}_2$	e)	$(\text{Na})\text{C}_{11}\text{H}_{18}\text{N}_2\text{O}_3$	f)	$(\text{C}_5\text{H}_5\text{N})_4\text{Cu}(\text{NCS})_4$
	180 g/mol		249 g/mol		728 g/mol

Calculate the mass of the following compounds.

a)	1.00 mol NaCl	b)	2.00 mol MgBr_2	c)	0.50 mol CCl_4
	58 g		368 g		76 g
d)	3.60 mol NaOH	e)	0.710 mol $\text{Ca}(\text{CN})_2$	f)	1.82 mol BaSO_4
	144 g		65.32 g		424.06 g

Calculate the number of moles of the following compounds.

a)	88 g CO_2	b)	198 g $\text{C}_2\text{H}_{12}\text{O}_6$	c)	65.25 g LiBr
	2.00 moles		1.5 moles		0.75 moles
a)	249 g NaOH	b)	878 g $\text{NaAlSi}_3\text{O}_8$	c)	7,327 g $\text{Cu}_2\text{FeSnS}_4$
	6.225 moles		3.35 moles		17.00 moles

24 Practice Problems Chemistry

**Russo Steve Silver Michael,Steve
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Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill,2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change

Techniques in Electroanalytical Chemistry Olja Simoska,Shelley D. Minteer,2022-06-07

Electrochemical science as a field is growing at a tremendous rate It was central to the emergence of chemistry as a discipline through the discovery of elements and is now poised to revolutionize energy neuroscience and organic synthesis among more traditional applications in corrosion prevention In this brief digital primer the authors introduce selected techniques in electroanalytical chemistry through text laboratory based tutorial videos and data analysis practice problems This primer is suitable for scientists interested in a brief introduction to the recent advances in electroanalytical chemistry instructors wanting to supplement an undergraduate or graduate course in instrumental analysis or the scientist wishing to incorporate electroanalytical techniques into projects to study reaction mechanisms

design energy conversion or energy storage devices and or design electrochemical sensors

General Chemistry Donald A. McQuarrie,Stanley Gill,2011-06-15

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Archibald Campbell Kennedy Smith,1968

Essential Laboratory Mathematics Catherine W. Johnson,Daniel L.

Timmons,Pamela E. Hall,2009-12-03 This hands on manual with pedagogical features that draw the learner into the content offers clear and complete coverage of the mathematical topics most often used in today s clinical and medical laboratories

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Partial Solutions Manual Darrell D. Ebbing, David Bookin, 1998 **Merrill Chemistry** Robert C.

Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Organic Chemistry* David R. Klein, 2017-08-14 In Organic Chemistry 3rd Edition Dr David Klein builds on the phenomenal success of the first two editions which presented his unique skills based approach to learning organic chemistry Dr Klein's skills based approach includes all of the concepts typically covered in an organic chemistry textbook and places special emphasis on skills development to support these concepts This emphasis on skills development in unique SkillBuilder examples provides extensive opportunities for two semester Organic Chemistry students to develop proficiency in the key skills necessary to succeed in organic chemistry Prentice Hall Chemistry, 2000

Problem Solving Guide and Workbook for Introductory Chemistry by Steve Russo, Mike Silver, Saundra Yancy

McGuire, Steve Russo, Mike Silver, 2002 Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills **Mathematics for the Clinical Laboratory**

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Symbolism and Calculations Stanley Wallace Morse, 1953 **Books of India**, 1964 *Essentials of Introductory Chemistry*
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