

WORKSHEET: BALANCING EQUATIONS

Balance the following equations by placing small whole-number coefficients in the appropriate places.

Synthesis

1. $\text{Ba} + \text{O}_2 \rightarrow \text{BaO}$
2. $\text{C} + \text{S} \rightarrow \text{CS}_2$
3. $\text{Li} + \text{O}_2 \rightarrow \text{Li}_2\text{O}$
4. $\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$
5. $\text{FeCl}_2 + \text{Cl}_2 \rightarrow \text{FeCl}_3$

Decomposition or Analysis

6. $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$
7. $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$
8. $\text{CuCO}_3 \rightarrow \text{CuO} + \text{CO}_2$
9. $\text{AuBr}_3 \rightarrow \text{Au} + \text{Br}_2$
10. $\text{UF}_4 \rightarrow \text{U} + \text{F}_2$

Single Replacement

11. $\text{Zn} + \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$
12. $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
13. $\text{Cu} + \text{AgNO}_3 \rightarrow \text{Cu}(\text{NO}_3)_2 + \text{Ag}$
14. $\text{K} + \text{H}_2\text{O} \rightarrow \text{KOH} + \text{H}_2$
15. $\text{Al} + \text{CuCl}_2 \rightarrow \text{AlCl}_3 + \text{Cu}$

Double Replacement

16. $\text{BaCl}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{NaCl} + \text{BaSO}_4$
17. $\text{ZnCl}_2 + (\text{NH}_4)_2\text{S} \rightarrow \text{NH}_4\text{Cl} + \text{ZnS}$
18. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
19. $\text{FeS} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2\text{S}$
20. $\text{AlCl}_3 + \text{NaOH} \rightarrow \text{Al}(\text{OH})_3 + \text{NaCl}$

Combustion

21. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
22. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
23. $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
24. $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
25. $\text{C}_5\text{H}_{12} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

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American Journal of Physics ,1993 **Nature** Sir Norman Lockyer, 1871 *Explanatory Models, Unit Standards, and*

Personalized Learning in Educational Measurement William P. Fisher Jr., Paula J. Massengill, 2022-10-15 The papers by Jack Stenner included in this book document the technical details of an art and science of measurement that creates new entrepreneurial business opportunities Jack brought theory instruments and data together in ways that are applicable not only in the context of a given test of reading or mathematics ability but which more importantly catalyzed literacy and numeracy capital in new fungible expressions Though Jack did not reflect in writing on the inferential constructive processes in which he engaged much can be learned by reviewing his work with his accomplishments in mind A Foreword by Stenner's colleague and co author on multiple works William P Fisher Jr provides key clues concerning a how Jack's understanding of measurement and its values aligns with social and historical studies of science and technology and b how recent developments in collaborations of psychometricians and metrologists are building on and expanding Jack's accomplishments This is an open access book

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 - Chapter 5: Conclusion
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- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of 233 Physical Science And Formulas Worksheet Answerkey.

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