



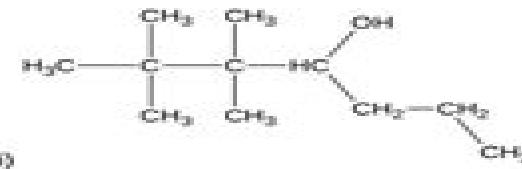
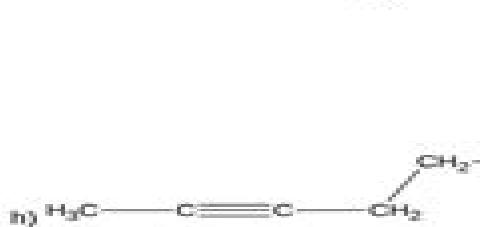
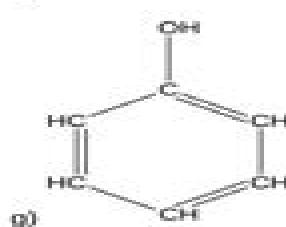
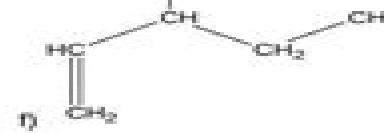
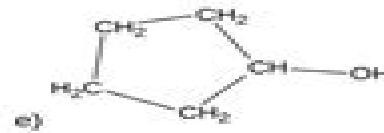
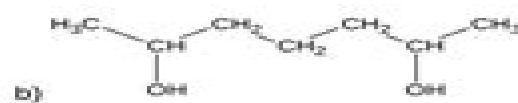
## 1. Draw the following alcohols

a) heptan-2-ol  
b) 3-methylhexan-1-ol  
c) cyclopropanol

d) 2,4,6-trichlorooctan-2-ol  
e) pentan-1,4-diol  
f) cyclohexene-1,3-diol

g) but-2-ene-1-ol  
h) 4-methylpent-2-yne-1-ol  
i) 3,4-dimethylcycloheptan-1-ol

## 2. Name the following alcohols



3. Explain why the propane that is used as a fuel in a BBQ is a gas at room temperature, but 2-propanol used as rubbing alcohol is a liquid at room temperature.

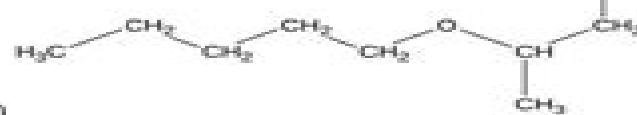
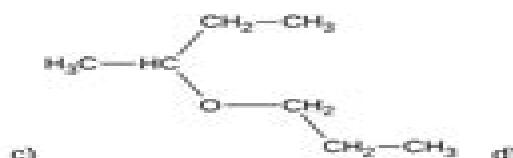
## 4. Draw the following ethers

a) 1-propoxypentane  
b) 2-ethoxybutane

c) 1-methoxy-4-chlorohexane  
d) 3-butoxy-2,4-dimethyloctane

e) 2-propoxyheptane  
f) 1-ethoxycyclopentane

## 5. Name the following ethers



# 22 Hydrocarbon Compounds Practice Problems Answer Key

**McGraw Hill**

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