

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate K_a and K_b for weak acids and bases, and pK_a and pK_b for weak acids and bases.
2. Calculate K_a and K_b for weak acids and bases, and pK_a and pK_b for weak acids and bases.
3. Calculate the pH of weak acid and base solutions.

Vocabulary

- | | | | |
|----------------------------|----------|----------------------------|----------|
| acid dissociation constant | 1. K_a | acid dissociation constant | 1. K_a |
| base dissociation constant | 2. K_b | base dissociation constant | 2. K_b |
| acid dissociation constant | 3. K_a | base dissociation constant | 3. K_b |
| base dissociation constant | 4. K_b | acid dissociation constant | 4. K_a |

Key Equations

- | | |
|------------------------------------|-------------------------------------|
| 1. $K_a = \frac{[H^+][A^-]}{[HA]}$ | 2. $K_b = \frac{[OH^-][HB^+]}{[B]}$ |
| 3. $pH = -\log[H^+]$ | 4. $pOH = -\log[OH^-]$ |

Part A Completion

Write the K_a expression for the acid dissociation constant for each of the following acids. Write the K_b expression for the base dissociation constant for each of the following bases.

1. HNO_3
2. HCl
3. H_2SO_4
4. H_2SO_3
5. H_2CO_3
6. H_2O
7. H_2O
8. H_2O
9. H_2O
10. H_2O
11. H_2O
12. H_2O
13. H_2O
14. H_2O
15. H_2O
16. H_2O
17. H_2O
18. H_2O
19. H_2O
20. H_2O

1. K_a
2. K_b
3. K_a
4. K_b
5. K_a
6. K_b
7. K_a
8. K_b
9. K_a
10. K_b
11. K_a
12. K_b
13. K_a
14. K_b
15. K_a
16. K_b
17. K_a
18. K_b
19. K_a
20. K_b

19 2 Section Review Hydrogen Ions And Acidity

Ying-Ying Zheng



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