

Types of Chemical Reactions Lab Answer Key

Chemical Reaction #1

Reactants (write in words)	hydrochloric acid and magnesium
Observations	accept most responses
Type of Reaction	single replacement
Predict Products (write in words)	magnesium chloride and hydrogen gas
Equation (write in symbols)	$2\text{HCl}_{\text{aq}} + \text{Mg}_{\text{ss}} \rightarrow \text{MgCl}_{\text{aq}} + \text{H}_{\text{g}}$

Chemical Reaction #2

Reactants (write in words)	copper metal and oxygen gas
Observations	accept most responses
Type of Reaction	synthesis
Predict Products (write in words)	copper oxide
Equation (write in symbols)	$2\text{Cu}_{\text{ss}} + \text{O}_{\text{g}} \rightarrow 2\text{CuO}_{\text{ss}}$

Chemical Reaction #3

Reactants (write in words)	methane gas and oxygen gas
Observations	accept most responses
Type of Reaction	combustion
Predict Products (write in words)	carbon dioxide and water
Equation (write in symbols)	$\text{CH}_{\text{4g}} + 2\text{O}_{\text{2g}} \rightarrow \text{CO}_{\text{2g}} + 2\text{H}_2\text{O}_{\text{2g}}$

Chemical Reaction #4

Reactants (write in words)	hydrogen peroxide
Observations	accept most responses
Type of Reaction	decomposition
Predict Products (write in words)	water and oxygen gas
Equation (write in symbols)	$2\text{H}_2\text{O}_{\text{2g}} \rightarrow 2\text{H}_2\text{O}_{\text{2g}} + \text{O}_{\text{2g}}$

Answer Key To Lab On Chemical Reactions

George Tyler Miller

Answer Key To Lab On Chemical Reactions:

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