

Name: \_\_\_\_\_

**ANSWER KEY**

Date: \_\_\_\_\_

Period: \_\_\_\_\_

## Radioactivity and Half-Lives Review Worksheet

1. What is a half-life?

The time it takes for half of the material you started with to decay.

2. If we start with 400 atoms of a radioactive substance, how many would remain after one half-life?

200 After 2 half-lives? 100 After 3 half-lives? 50 After 4 half-lives? 25

3. If we start with 16 grams of a substance, how much will remain after 3 half-lives?

$16 \rightarrow 8 \rightarrow 4 \rightarrow 2$  2 grams after 3 half-lives.

4. A sample of carbon-14 has been decaying for 22,920 years and is now 35.0 grams. What was the size of the original sample? (The half-life of carbon-14 is 5,730 years)

$22920/5730 = 4$  half lives You will need to work backwards

$35 \rightarrow 70 \rightarrow 140 \rightarrow 280 \rightarrow 560$  grams

7. How old is a skeleton sample if the current amount of carbon in the bones is 3.125%? (Assume you started with 100%)

$100\% \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow 6.25 \rightarrow 3.125$  5 half lives  $\times 5,730 = 28,650$  years

8. If you started with 120 grams of a radioactive substance, and now you have 15 grams, how many half-lives have passed?

$120 \rightarrow 60 \rightarrow 30 \rightarrow 15$  3 half-lives

9. The half-life of isotope X is 2.0 years. How many years would it take for a 4.0 mg sample of isotope X to decay and have only 0.50 mg of it remain?

$4.0 \rightarrow 2.0 \rightarrow 1.0 \rightarrow 0.50$  3 half lives  $\times 2$  years = 6.0 years

10. The half-life of Po-218 is three minutes. How much of a 200 atom sample will remain after 15 minutes?

$15 \text{ minutes}/3 = 5$  half lives  $200 \rightarrow 100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow 6.25$  atoms

11. The half-life of Au-198 is 2.69 days. How long does it take a 180g sample to decay to 1/8 its original mass?

$1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8}$  3 half lives  $\times 2.69$  days = 8.07 days

12. What is the half-life of a radioactive sample if 100.0 grams of it decays to 12.5 grams in 24.3 hours?

$100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5$   $24.3/3$  half lives = 8.10 hours

13. How long will it take a sample of  $^{131}\text{I}$  to decay to 12.5% assuming its half-life is 8.07 days?

# Answer Key To Half Lives

**Ying Liu**

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