

Getting to Know the Elements!

Use your Periodic Table and note pages to help you get to know these elements a little better! Remember...A.P.E. (A.N.S.)

1. How many protons does Calcium have? _____
2. How many electrons does Yttrium have? _____
3. How many neutrons does Titanium have? _____
4. The subatomic particle that has a neutral charge is the: _____
5. The subatomic particle that has a negative charge is the: _____
6. The subatomic particle that has a positive charge is the: _____
7. What element has 13 protons? _____
8. What element has 21 neutrons? _____
9. What element has 0 neutrons? _____
10. How many electrons does O₂ have? _____
11. What is the sum of the atomic masses of B₁ and B₇? _____
12. How many neutrons does Gold have? _____
13. What is the sum of the number of protons of Iridium and the number of protons in Strontium? _____
14. What is the element symbol for Platinum? _____
15. What is the difference between the number of electrons in an atom of Tungsten and the electrons in an atom of Tin? _____
16. What particles are found in the nucleus of an atom? _____ and _____
17. Where do you find the electrons in an atom? _____
18. How many electrons can the 2nd electron cloud hold? _____

Answer Key I Getting To Know

Richard L. Strauss

Answer Key I Getting To Know:

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