



MIXTURES AND SOLUTIONS

NGSS - GRADE 7 - SCIENCE

Mixtures and Solutions

Total: 30 marks

Reading Passage

Read the passage carefully before answering the questions that follow.

Mixtures are combinations of two or more substances that retain their own properties. Mixtures can be classified into homogeneous and heterogeneous mixtures. Homogeneous mixtures have uniform composition throughout. An example is saltwater, as nearly as every part has a clear transparent mixture. Heterogeneous mixtures bring a contrast of vastly different substances that can often be separated.

In addition, particles are at the molecular or ionic level, making them homogeneous. Colloids are mixtures with larger particles that may remain in suspension after light is scattered. Separations contain large particles such as mostly water or oil and water emulsion. These distinctions are how easily the mixtures can be separated.

Understanding solubility is essential in characterizing mixtures. Solubility is amount of a solute that can dissolve in a solvent of a given temperature. In addition, when we have solute can dissolve. The mixture is saturated. For cooling point elevation or freezing point depression depend on the number amount and their identity, affecting mixture behavior.

Techniques like filtration, separation, and distillation are employed to isolate their properties. For example, sand can be separated from water by separation can recover dissolved salts. The study of mixtures not only has properties but also highlights practical applications such as creating alloy purifying drinking water.

True or False Statements

Identify whether each statement is True or False.

- Homogeneous mixtures have uniform composition throughout.
- Separations are mixtures where particles remain suspended and never settle.
- Colloids scatter light and show the Tyndall effect.
- Separation occurs when no more solute can dissolve in a solvent at a given temperature.
- Distillation is an effective method to separate dissolved salts from water.
- Colloids have particles larger than those in solutions but smaller than in suspensions.
- Separation can be used to recover salt from a saltwater solution.
- Homogeneous mixtures are homogeneous because they do not have any solutes.
- Particle size determines whether a mixture is a solution, colloid, or suspension.

- All mixtures can be separated by physical means.

- True
- False

Multiple Choice Questions

Select the answer that best completes the statement.

- Which of the following is an example of a homogeneous mixture?

- Sand and gravel
- Saltwater
- Oil and water
- Heated river water

- What property of water?

- Surface area
- Color
- Density
- Volume

- Which technique is used for separating a solid from a liquid?

- Filtration
- Distillation
- Chromatography
- Decanting

- After a solution reaches maximum solute in a given amount of solvent, what is the solution?

- Soluble solution
- Super-saturated solution
- Unsaturated solution
- Saturated solution

- Which property depends on the number of solute particles and their size?

- Mass
- Colligative properties
- Color



Answer Key For Solutions And Other Mixtures

**Gautam Allahbadia, Masashige
Kuwayama, Goral Gandhi**

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