

CHAPTER 4 PRACTICE TEST: SOLUTION STOICHIOMETRY

1) Which of the following are weak electrolytes? HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KCl
 A) HCl, KCl
 B) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KCl
 C) $\text{HC}_2\text{H}_3\text{O}_2$, KCl
 D) $\text{HC}_2\text{H}_3\text{O}_2$, NH_3
 E) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, KCl

Answer: D Diff: 2

2) What are the spectator ions in the reaction between KOH (aq) and HNO_3 (aq)?

A) K^+ and H^+ C) K^+ and NO_3^- E) OH^- only
 B) H^+ and OH^- D) H^+ and NO_3^-

Answer: C Diff: 2

3) The balanced net ionic equation for precipitation of CaCO_3 when aqueous solutions of Na_2CO_3 and CaCl_2 are mixed is _____.

A) 2Na^+ (aq) + CO_3^{2-} (aq) \rightarrow Na_2CO_3 (aq)
 B) 2Na^+ (aq) + 2Cl^- (aq) \rightarrow 2NaCl (aq)
 C) Na^+ (aq) + Cl^- (aq) \rightarrow NaCl (aq)
 D) Ca^{2+} (aq) + CO_3^{2-} (aq) \rightarrow CaCO_3 (s)

Answer: D Diff: 3

4) Which of the following are strong acids? HI, HNO_3 , HF, HBr

A) HF, HBr C) HI, HF, HBr E) HI, HNO_3 , HBr
 B) HI, HNO_3 , HF, HBr D) HNO_3 , HF, HBr

Answer: E Diff: 2

5) A neutralization reaction between an acid and a metal hydroxide produces _____.

A) water and a salt C) oxygen gas E) ammonia
 B) hydrogen gas D) sodium hydroxide

Answer: A Diff: 2

6) Which of these metals is the least easily oxidized? Na, Au, Fe, Ca, Ag

A) Na B) Au C) Fe D) Ca E) Ag

Answer: B Diff: 3

7) Based on the equations below, which metal is the most active?



A) Ni B) Ag C) Cu D) Pb E) N

Answer: A Diff: 3

8) What is the concentration (M) of KCl in a solution made by mixing 25.0 mL of 0.100 M KCl with 50.0 mL of 0.100 M KCl?

A) 0.100 B) 0.0500 C) 0.0333 D) 0.0250 E) 125

Answer: A Diff: 3

Answer Key For Module 13 Chemistry Test

Johann Gasteiger

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Table of Contents Answer Key For Module 13 Chemistry Test

1. Understanding the eBook Answer Key For Module 13 Chemistry Test
 - The Rise of Digital Reading Answer Key For Module 13 Chemistry Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Module 13 Chemistry Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Module 13 Chemistry Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Module 13 Chemistry Test
 - Personalized Recommendations
 - Answer Key For Module 13 Chemistry Test User Reviews and Ratings
 - Answer Key For Module 13 Chemistry Test and Bestseller Lists

5. Accessing Answer Key For Module 13 Chemistry Test Free and Paid eBooks
 - Answer Key For Module 13 Chemistry Test Public Domain eBooks
 - Answer Key For Module 13 Chemistry Test eBook Subscription Services
 - Answer Key For Module 13 Chemistry Test Budget-Friendly Options
6. Navigating Answer Key For Module 13 Chemistry Test eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answer Key For Module 13 Chemistry Test Compatibility with Devices
 - Answer Key For Module 13 Chemistry Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Module 13 Chemistry Test
 - Highlighting and Note-Taking Answer Key For Module 13 Chemistry Test
 - Interactive Elements Answer Key For Module 13 Chemistry Test
8. Staying Engaged with Answer Key For Module 13 Chemistry Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Module 13 Chemistry Test
9. Balancing eBooks and Physical Books Answer Key For Module 13 Chemistry Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Module 13 Chemistry Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key For Module 13 Chemistry Test
 - Setting Reading Goals Answer Key For Module 13 Chemistry Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key For Module 13 Chemistry Test
 - Fact-Checking eBook Content of Answer Key For Module 13 Chemistry Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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