

General Chemistry – Unit 1 Worksheet 6**Dimensional Analysis**

Use the factor-label method to make the following conversions. Remember to use the appropriate number of s.f's in your answer.

Part 1

1. $74 \text{ cm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ meters}$

2. $8.32 \times 10^{-2} \text{ kg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ grams}$

3. $55.5 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ cm}^3$

4. $0.00527 \text{ cal} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ kilocalories}$

5. $9.52 \times 10^{-4} \text{ m} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ micrometers}$

6. $41.0 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ liters}$

7. $6.0 \times 10^{-1} \text{ g} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ mg}$

8. $8.34 \times 10^{-9} \text{ cg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ g}$

9. $5.0 \times 10^3 \text{ mm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ m}$

10. $1 \text{ day} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ seconds}$

11. $5 \times 10^4 \text{ mm} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ km}$

12. $9.1 \times 10^{-13} \text{ kg} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ mg}$

13. $1 \text{ year} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ hours (approximately)}$

Answer Key Chemistry Unit 6

Marion F. Gooding

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