



Balancing Chemical Equations

Directions: Balance each of the following chemical equations.

- 1) $\underline{\quad}$ H_3PO_4 + $\underline{\quad}$ KOH \rightarrow $\underline{\quad}$ K_3PO_4 + $\underline{\quad}$ H_2O
- 2) $\underline{\quad}$ K + $\underline{\quad}$ B_2O_3 \rightarrow $\underline{\quad}$ K_2O + $\underline{\quad}$ B
- 3) $\underline{\quad}$ HCl + $\underline{\quad}$ NaOH \rightarrow $\underline{\quad}$ NaCl + $\underline{\quad}$ H_2O
- 4) $\underline{\quad}$ Na + $\underline{\quad}$ NaNO_3 \rightarrow $\underline{\quad}$ Na_2O + $\underline{\quad}$ N_2
- 5) $\underline{\quad}$ C + $\underline{\quad}$ S_8 \rightarrow $\underline{\quad}$ CS_2
- 6) $\underline{\quad}$ Na + $\underline{\quad}$ O_2 \rightarrow $\underline{\quad}$ Na_2O
- 7) $\underline{\quad}$ N_2 + $\underline{\quad}$ O_2 \rightarrow $\underline{\quad}$ N_2O_5
- 8) $\underline{\quad}$ H_3PO_4 + $\underline{\quad}$ $\text{Mg}(\text{OH})_2$ \rightarrow $\underline{\quad}$ $\text{Mg}_3(\text{PO}_4)_2$ + $\underline{\quad}$ H_2O
- 9) $\underline{\quad}$ NaOH + $\underline{\quad}$ H_2CO_3 \rightarrow $\underline{\quad}$ Na_2CO_3 + $\underline{\quad}$ H_2O
- 10) $\underline{\quad}$ KOH + $\underline{\quad}$ HBr \rightarrow $\underline{\quad}$ KBr + $\underline{\quad}$ H_2O
- 11) $\underline{\quad}$ Na + $\underline{\quad}$ O_2 \rightarrow $\underline{\quad}$ Na_2O
- 12) $\underline{\quad}$ $\text{Al}(\text{OH})_3$ + $\underline{\quad}$ H_2CO_3 \rightarrow $\underline{\quad}$ $\text{Al}_2(\text{CO}_3)_3$ + $\underline{\quad}$ H_2O
- 13) $\underline{\quad}$ Al + $\underline{\quad}$ S_8 \rightarrow $\underline{\quad}$ Al_2S_3
- 14) $\underline{\quad}$ Cs + $\underline{\quad}$ N_2 \rightarrow $\underline{\quad}$ Cs_2N
- 15) $\underline{\quad}$ Mg + $\underline{\quad}$ Cl_2 \rightarrow $\underline{\quad}$ MgCl_2
- 16) $\underline{\quad}$ Rb + $\underline{\quad}$ RbNO_3 \rightarrow $\underline{\quad}$ Rb_2O + $\underline{\quad}$ N_2
- 17) $\underline{\quad}$ C_6H_6 + $\underline{\quad}$ O_2 \rightarrow $\underline{\quad}$ CO_2 + $\underline{\quad}$ H_2O
- 18) $\underline{\quad}$ N_2 + $\underline{\quad}$ H_2 \rightarrow $\underline{\quad}$ NH_3

Another Balancing Chemical Equations Worksheet

Answers

SJ Ball

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