

30. A scientific law is established only after
- discoveries have been made.
 - experiments have been carried out and proven true.
 - hypothesis has been tested.
 - reasonable explanations have been put forward.
 - theory has been extensively tested and proven true.
31. The following are properties of group VI elements **except** that their
- atomic radius increases down the group.
 - electronegativity decreases down the group.
 - melting and boiling points increase down the group.
 - metallic properties decrease down the group.
 - outer most shell has six electrons
32. What volume in dm^3 of water will be added to 1dm^3 of 6mol dm^{-3} solution to make it 2mol dm^{-3} ?
- 0.2
 - 0.3
 - 2.0
 - 3.0
 - 12.0
33. In the periodic table, elements in the same period show
- progressive decrease in electronegativity.
 - progressive increase in atomic number.
 - similarities in chemical properties.
 - similarities in electronic structures.
 - similarities in physical properties.
34. If the rate of diffusion of oxygen is $7.5\text{ cm}^3/\text{s}$ and the time taken to diffuse is 12 seconds, calculate the volume in (cm^3).
- 80
 - 89
 - 90
 - 91
 - 99
35. What similarity exists between a chlorine atom and chlorine ion?
- Chemical property
 - Electron charge
 - Electronic configuration
 - Number of electrons
 - Number of protons
36. 28.4 g of a salt dissolved in 250 cm^3 of distilled water to give a concentration of 0.8mol dm^{-3} . Calculate the molar mass of the salt.
- 35.5
 - 71.0
 - 142.0
 - 250.0
 - 278.4

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X.D. Xu, Bin Li, Q.M. Lu, X.Y. Yan, J.L. Li



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