

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1  
☒ 2  
☐ 3  
☐ 4

2. At constant  $T_1 = 100^\circ\text{C}$ , what is the molar volume of an ideal gas?
- ☒ 22.4  
☐ 22.7  
☐ 22.8  
☐ 22.9

3. The universal gas constant is the same as:
- ☐ Boltzmann's constant  
☐ gas constant  
☒ both of these  
☐ none of these

4. For constant  $T_1 = 100^\circ\text{C}$ ,  $n = 1.0$  mol, how many moles of gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1.0 mol  
☐ 2.0 mol  
☐ 3.0 mol  
☒ 4.0 mol

5. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K  
☒ 8.314 J/mol·K  
☐ 8.314 J/mol·K  
☐ 8.314 J/mol·K

6. What is the value of the universal gas constant?
- ☐ 8.314 J/mol·K  
☐ 8.314 J/mol·K  
☒ 8.314 J/mol·K  
☐ 8.314 J/mol·K

7. A cylinder of compressed gas is at 100 atm. The gas is at 100 atm. The gas is at 100 atm.
- ☒ 100 atm  
☐ 100 atm  
☐ 100 atm  
☐ 100 atm

8. For constant  $T_1 = 100^\circ\text{C}$ ,  $n = 1.0$  mol, how many moles of gas are in a cylinder at 100 atm pressure, 20°C?
- ☒ 1.0 mol  
☐ 2.0 mol  
☐ 3.0 mol  
☐ 4.0 mol

9. At constant  $T_1 = 100^\circ\text{C}$ ,  $n = 1.0$  mol, what is the molar volume of an ideal gas?
- ☐ 22.4  
☐ 22.7  
☒ 22.8  
☐ 22.9

10. The universal gas constant is the same as:
- ☐ Boltzmann's constant  
☐ gas constant  
☒ both of these  
☐ none of these

11. The universal gas constant is the same as:
- ☐ Boltzmann's constant  
☐ gas constant  
☒ both of these  
☐ none of these

# Multiple Choice Stoichiometry Test

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countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems

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