

Chemistry - Review Chapters (45) 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE: 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE: 18

5. State the number of valence electrons in an atom of:

a. sulfur
6

b. calcium
2

c. chlorine
7

d. arsenic
5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10
Ne, NEON

b. 18
Ar, ARGON

c. 36
Kr, KRYPTON

d. 90
Th, THORIUM

7. List, by number, both the period and group of each of these elements.

Symbol	Period	Group
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a. beryllium

Be

2

2

b. iron

Fe

4

8

c. lead

Pb

6

14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl

b. Na and Li

c. Na and Cu

d. Na and He

9. Which of the following pairs of elements belong to the same group?

a. H and He

b. Li and Be

c. C and Pb

d. Os and Ge

10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

F ORBITALS

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Table of Contents Answer Key To Prentice Hall Chemistry Work

1. Understanding the eBook Answer Key To Prentice Hall Chemistry Work
 - The Rise of Digital Reading Answer Key To Prentice Hall Chemistry Work
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Prentice Hall Chemistry Work
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Prentice Hall Chemistry Work
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Prentice Hall Chemistry Work
 - Personalized Recommendations
 - Answer Key To Prentice Hall Chemistry Work User Reviews and Ratings
 - Answer Key To Prentice Hall Chemistry Work and Bestseller Lists

5. Accessing Answer Key To Prentice Hall Chemistry Work Free and Paid eBooks
 - Answer Key To Prentice Hall Chemistry Work Public Domain eBooks
 - Answer Key To Prentice Hall Chemistry Work eBook Subscription Services
 - Answer Key To Prentice Hall Chemistry Work Budget-Friendly Options
6. Navigating Answer Key To Prentice Hall Chemistry Work eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answer Key To Prentice Hall Chemistry Work Compatibility with Devices
 - Answer Key To Prentice Hall Chemistry Work Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Prentice Hall Chemistry Work
 - Highlighting and Note-Taking Answer Key To Prentice Hall Chemistry Work
 - Interactive Elements Answer Key To Prentice Hall Chemistry Work
8. Staying Engaged with Answer Key To Prentice Hall Chemistry Work
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Prentice Hall Chemistry Work
9. Balancing eBooks and Physical Books Answer Key To Prentice Hall Chemistry Work
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Prentice Hall Chemistry Work
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key To Prentice Hall Chemistry Work
 - Setting Reading Goals Answer Key To Prentice Hall Chemistry Work
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key To Prentice Hall Chemistry Work
 - Fact-Checking eBook Content of Answer Key To Prentice Hall Chemistry Work
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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