

Name: _____
 Date: _____

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Speed and Acceleration

Speed

Speed is the rate of change in position. Speed is the distance traveled divided by the time taken to travel that distance.

$$\text{Speed} = \frac{\text{Distance}}{\text{Time}}$$

Speed is a scalar quantity. It has magnitude but no direction.



A car traveling at 60 km/h is moving faster than a car traveling at 30 km/h. But is it moving faster in the same direction?

A car traveling at 60 km/h is moving faster than a car traveling at 30 km/h.		A car traveling at 60 km/h is moving faster than a car traveling at 30 km/h.		A car traveling at 60 km/h is moving faster than a car traveling at 30 km/h.	
Speed	60 km/h	Speed	60 km/h	Speed	60 km/h
Direction	North	Direction	North	Direction	North
Speed	30 km/h	Speed	30 km/h	Speed	30 km/h
Direction	North	Direction	North	Direction	North

Acceleration

Acceleration is the rate of change in velocity. It is a vector quantity. It has magnitude and direction.

Acceleration is the rate of change in velocity. It is a vector quantity. It has magnitude and direction.



Acceleration is the rate of change in velocity. It is a vector quantity. It has magnitude and direction.

Velocity

Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Time	Position
0	0
1	10
2	20
3	30
4	40
5	50

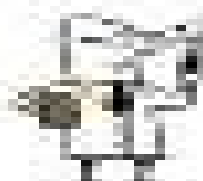
Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Velocity is the rate of change in position. It is a vector quantity. It has magnitude and direction.

Position

Position is the location of an object. It is a vector quantity. It has magnitude and direction.

Position is the location of an object. It is a vector quantity. It has magnitude and direction.



Acceleration

Acceleration is the rate of change in velocity. It is a vector quantity. It has magnitude and direction.

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Acceleration is the rate of change in velocity. It is a vector quantity. It has magnitude and direction.

Answer Key For Speed Experiments Murray

Michael Seilmaier



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes

chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between

temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of

declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95

82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts

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