



## Study Guide for Chemistry Midterm Exam

### Midterm Exam Facts

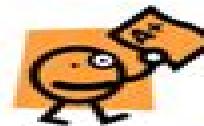
- The midterm exam is scheduled for \_\_\_\_\_ . You will have 90 minutes.
- The midterm exam is cumulative. This means it covers everything we have studied in chemistry this semester.
- The exam consists of multiple choice questions, fill in the chart questions, and several short answer questions.

### How Does the Midterm Exam Affect My Grade?

- The midterm exam counts 15% of your overall Quarter grade for chemistry.
- Students historically score within 10% of their test average on the midterm exam. So, if Polly Perfect's test average is an 85%, she can expect to score 75-95% on the midterm – if she studies!

### What Should I Review/Study?

- Unit Tests and Quizzes
- Previous study guides
- Online review resources (both on the midterm page and the unit pages)
- This Study Guide & Answer Key



### Which Reference Sheets Can I Use on the Midterm?

- 4" x 6" notecard (hand-written front & back)
- Periodic Table (Miss Scott's class set)
- Unit Conversion Factors Reference Sheet (i.e. 1 inch = 2.54 cm)
- Average Atomic Mass Reference Sheet
- Periodic Table Flipbook

### What if I Need Extra Help?

- It is much better to study a little every day than to leave things until the last minute. Seek help early.
- Find a study buddy/friend/significant other ☺ who is in chemistry and study together.
- Consult Miss Scott's weekly schedule. She is happy to help you if you come prepared with specific questions. Arriving unprepared and asking "What should I know for the midterm exam?" is both unhelpful and unwise.

### Last Piece of Advice

Chemistry is a challenging college preparatory class. Recognize that your college/university instructors will not provide you with a study guide that covers "everything" that will be on their exams. In fact, most professors won't provide you with a study guide at all. You can thank Miss Scott for putting together this study guide for you. Please view this as a "guide," and understand it will not tell you everything that is on the midterm exam. However, if you are able to answer all of the questions successfully in this guide, and you truly understand all of the bulleted concepts listed for you to study, your chances of success are very good.

### Units 1 – 3 – Measurements, Scientific Method, Math, & Matter

1. Identify each of the following as an example of a physical (P) change or a chemical (C) change:
  - a. the baking of bread **C**
  - b. the melting of iron **P**
  - c. the dissolving of sugar in water **P**
  - d. tarnishing of silver **C**
2. Matter – Different types and Classification Diagram (flowchart)
3. Scientific Method
4. Perform the calculations and round off each answer to the correct number of significant figures.
 

a. $162.1g + 38.73g + 1.554g = 202.4 \text{ g}$	b. $21.9\text{m} - 6.34\text{m} + 15.7\text{m} = 173 \text{ m}$
c. $9.88\text{cm} \times 7.2 \text{ cm} = 71 \text{ cm}^2$	d. $44.75 \text{ g} / 1.62 \text{ L} = 27.6 \text{ g/L}$
5. Express the following quantities in scientific notation
 

a. $600$	$6 \times 10^2$	b. $7770$	$7.77 \times 10^3$	c. $0.125$	$1.25 \times 10^{-2}$	d. $0.0000025$	$2.5 \times 10^{-6}$
----------	-----------------	-----------	--------------------	------------	-----------------------	----------------	----------------------
6. At a track meet, you time a friend running 100 m at 11.00 seconds. The official time of her race is 10.87 seconds. What is your % error? **1.196%**
7. Express each of the following as a whole number or decimal (not in scientific notation).
 

a. $2.5 \times 10^3$	<b>2500</b>	b. $6.25 \times 10^{-6}$	<b>0.00000625</b>
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# Answer Key Chemistry Midterm Exam Study Guide

**Michael Hoelscher**

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## **Table of Contents Answer Key Chemistry Midterm Exam Study Guide**

1. Understanding the eBook Answer Key Chemistry Midterm Exam Study Guide
  - The Rise of Digital Reading Answer Key Chemistry Midterm Exam Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Chemistry Midterm Exam Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key Chemistry Midterm Exam Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Chemistry Midterm Exam Study Guide
  - Personalized Recommendations
  - Answer Key Chemistry Midterm Exam Study Guide User Reviews and Ratings
  - Answer Key Chemistry Midterm Exam Study Guide and Bestseller Lists

5. Accessing Answer Key Chemistry Midterm Exam Study Guide Free and Paid eBooks
  - Answer Key Chemistry Midterm Exam Study Guide Public Domain eBooks
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  - Answer Key Chemistry Midterm Exam Study Guide Budget-Friendly Options
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  - Answer Key Chemistry Midterm Exam Study Guide Compatibility with Devices
  - Answer Key Chemistry Midterm Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key Chemistry Midterm Exam Study Guide
  - Highlighting and Note-Taking Answer Key Chemistry Midterm Exam Study Guide
  - Interactive Elements Answer Key Chemistry Midterm Exam Study Guide
8. Staying Engaged with Answer Key Chemistry Midterm Exam Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key Chemistry Midterm Exam Study Guide
9. Balancing eBooks and Physical Books Answer Key Chemistry Midterm Exam Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key Chemistry Midterm Exam Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Answer Key Chemistry Midterm Exam Study Guide
  - Setting Reading Goals Answer Key Chemistry Midterm Exam Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key Chemistry Midterm Exam Study Guide
  - Fact-Checking eBook Content of Answer Key Chemistry Midterm Exam Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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