

## Section Review 8.1

### Part A Completion

1. compound	6. low
2. molecular	7. high
3. nonmetals	8. atoms
4. diatomic	9. structure
5. molecular formula	

### Part B True-False

10. ST	12. AT	14. ST
11. NT	13. NT	

### Part C Matching

15. e	17. d	19. c
16. a	18. b	

### Part D Questions and Problems

20. molecular compound  
21. a. 4 carbon atoms, 10 hydrogen atoms  
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom  
22. a. molecule      d. atom  
b. atom      e. molecule  
c. molecule

## Section Review 8.2

### Part A Completion

- stable electron
- covalent
- shared
- single
- unshared pairs
- double/triple
- coordinate covalent bond
- Energy
- bond dissociation energy
- resonance structure

### Part B True-False

11. NT	13. AT	15. ST
12. NT	14. AT	16. NT

### Part C Matching

17. e	19. b	21. d
18. a	20. c	

## Part D Questions and Problems

22. a.  $\text{:Br}-\text{Br}:$   
b.  $\text{H}-\text{C}=\text{C}-\text{H}$   
c.  $\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{N}-\text{H} \\ | \\ \text{H} \end{array}$

## Section 8.3

### Part A Completion

- molecular orbitals
- bonding orbital
- lower
- sigma or  $\sigma$
- pi or  $\pi$
- three-dimensional
- VSEPR theory
- orbital hybridization

### Part B True-False

9. AT	11. NT	13. ST
10. ST	12. AT	14. AT

### Part C Matching

15. d	17. a	19. c
16. e	18. b	

### Part D Questions and Problems

20. Reading from left to right:  
 $\text{sp}^3$   $\text{sp}^2$   $\text{sp}^2$   $\text{sp}$   $\text{sp}$   $\text{sp}^2$

## Section 8.4

### Part A Completion

- equally
- nons polar
- unequally
- polar
- electronegativities
- dipole interactions
- hydrogen bond
- electronegative
- oxygen, nitrogen, or fluorine

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