

Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?
Its overall size and a balance between protons and neutrons.
- b. How does an unstable atom gain stability?
It undergoes radioactive decay.
- c. What part of an atom undergoes change during radioactive decay?
The nucleus.

2. Which of the human senses can detect radiation?

Our eyes can detect colors, but we have no sense to detect alpha, beta, or gamma radiation.

3. Explain the difference between an isotope and a radioactive isotope.

An isotope has a different set of protons and neutrons. Radioactive isotope is large unstable, and has a short life.

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	Four protons	Positive (+)	Stopped by skin or paper
Beta	One Neutron (isotrons)	Negative (-)	Stopped by sheet of metal or thick paper
Gamma	One Neutron	Neutral	Stopped by several inches of lead or lead of concrete

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	+1
Gamma	No change	No change

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



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Table of Contents Answer Key To Nuclear Radiation

1. Understanding the eBook Answer Key To Nuclear Radiation
 - The Rise of Digital Reading Answer Key To Nuclear Radiation
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Nuclear Radiation
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Nuclear Radiation
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Nuclear Radiation
 - Personalized Recommendations
 - Answer Key To Nuclear Radiation User Reviews and Ratings

- Answer Key To Nuclear Radiation and Bestseller Lists
- 5. Accessing Answer Key To Nuclear Radiation Free and Paid eBooks
 - Answer Key To Nuclear Radiation Public Domain eBooks
 - Answer Key To Nuclear Radiation eBook Subscription Services
 - Answer Key To Nuclear Radiation Budget-Friendly Options
- 6. Navigating Answer Key To Nuclear Radiation eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answer Key To Nuclear Radiation Compatibility with Devices
 - Answer Key To Nuclear Radiation Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Nuclear Radiation
 - Highlighting and Note-Taking Answer Key To Nuclear Radiation
 - Interactive Elements Answer Key To Nuclear Radiation
- 8. Staying Engaged with Answer Key To Nuclear Radiation
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Nuclear Radiation
- 9. Balancing eBooks and Physical Books Answer Key To Nuclear Radiation
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Nuclear Radiation
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Nuclear Radiation
 - Setting Reading Goals Answer Key To Nuclear Radiation
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Nuclear Radiation
 - Fact-Checking eBook Content of Answer Key To Nuclear Radiation
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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