

Mole Conversions Worksheet #1

1. Mole \rightarrow Mass Conversions – using molar mass of each substance, convert the following quantities.

- | | |
|------------------------------------------------|-----------------------------------------------------------------------------------------------|
| a. 10.0 mol Cr 520 g | f. 0.160 mol H_2O 2.88 g |
| b. 3.32 mol K 130 g | g. 5.08 mol $\text{Ca}(\text{NO}_3)_2$ 834 g |
| c. 2.20×10^{-5} mol Sn 0.261 g | h. 15.0 mol H_2SO_4 1470 g |
| d. 0.720 mol Be 6.48 g | i. 4.52×10^{-5} mol C_2H_6 1.27×10^{-3} g |
| e. 2.40 mol N_2 67.2 g | j. 0.0112 mol K_2CO_3 1.55 g |

2. Mass \rightarrow Mole Conversions – using molar mass of each substance convert the following quantities.

- | | |
|---------------------------------------------------------------------------|-------------------------------------------------------------------------------------------------|
| a. 72.0 g Ar 1.80 mol | f. 27.4 g NO_2 0.596 mol |
| b. 3.70×10^{-3} g B 3.43×10^{-2} mol | g. 5.00 g H_2 2.50 mol |
| c. 187 g Al 6.93 mol | h. 2.64×10^{-4} g Li_3PO_4 2.28×10^{-6} mol |
| d. 333 g SnF_2 2.13 mol | i. 11.0 g CH_4 0.688 mol |
| e. 7.21×10^{-2} g He 1.80×10^{-2} mol | j. 847 g $(\text{NH}_4)_2\text{CO}_3$ 8.82 mol |

3. What is the volume of the following gases?

- a. 5.40 mol O_2 **121 L**
b. 3.20×10^{-2} mol CO_2 **0.717 L**
c. 0.960 mol SO_3 **21.5 L**

4. How many moles are in each of the following volumes?

- a. 89.6 L Ne **4.00 mol**
b. 1.00×10^3 L C_2H_6 **44.6 mol**
c. 5.42×10^{-1} F_2 **2.42×10^{-2} mol**

5. Find the number of moles in each of the number of representative particles.

- a. 1.20×10^{25} atoms of P **19.9 mol**
b. 3.87×10^{23} molecules of AlF_3 **6.45×10^{-3} mol**
c. 4.81×10^{24} molecules of NH_3 **7.99×10^{-3} mol**

6. How many representative particles are in each of the following mole quantities?

- a. 1.24 mol Cl_2 **7.46×10^{23} molecules**
b. 4.20×10^{-5} mol K_2S **2.53×10^{23} molecules**
c. 34.02 mol $\text{Ca}(\text{OH})_2$ **2.048×10^{25} molecules**

7. Convert the following two-step quantities, converting first to moles and then to the desired quantity.

- a. Find the number of molecules in 60.0 g of N_2O . **8.21×10^{23} molecules**
b. Find the volume of 3.24×10^{22} molecules of Ne. **1.21 L**
c. Find the mass of 18.0 L of CH_4 . **12.9 g**
d. Find the volume of 835 g of SO_3 . **234 L**
e. Find the mass of one atom of nickel. **1×10^{-22} g**

Answer Key Mass Mole Conversion Worksheet

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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