

# OCEANOGRAPHY UNIT

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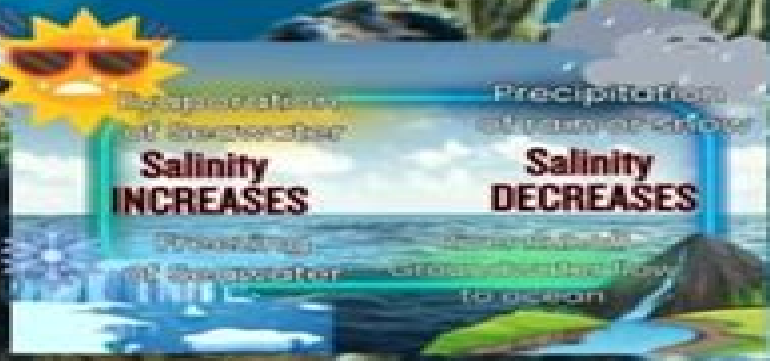
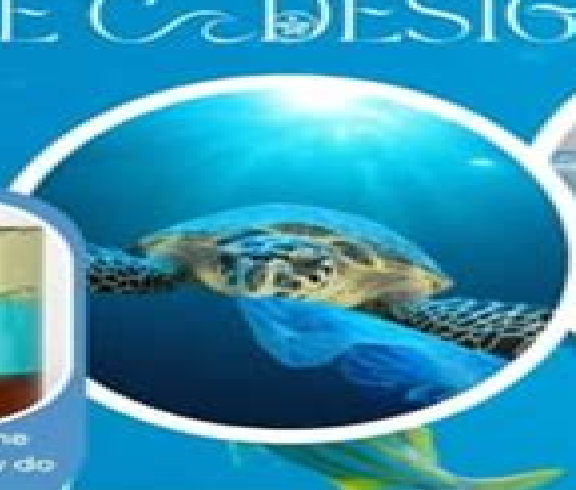
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# Answer Key Investigation Manual Oceanography

**JL Elias**



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web the rate determining slower step gives a rate law showing second order dependence on the  $[a]^2$  concentration and the sum of the two elementary equations gives the overall net reaction in general when the rate determining slower step is the first step in the reaction mechanism the rate law for the overall reaction is the same as the rate

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web rate laws and reaction order the relation between the rate of a reaction and the concentrations of reactants is expressed



by its rate law for example the rate of the gas phase decomposition of dinitrogen pentoxide  $2\text{N}_2\text{O}_5 \rightarrow 2\text{N}_2 + 5\text{O}_2$  has been found to be directly proportional to the concentration of  $\text{N}_2\text{O}_5$  text rate  $k$   $\text{N}$

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web the rate determining step limits the overall rate and therefore determines the rate law for the overall reaction

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide gas and hydrogen gas to form nitrogen gas and water vapor the following data were collected for this reaction at 1280 text o text c

see table below notice that the starting concentrations of  $\text{CeNO}$  and  $\text{CeH}$

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