

# CHEMISTRY: COUNTING ATOMS IN COMPOUNDS WORKSHEET #7.0.1

**INSTRUCTIONS:** Write the quantity of atoms of each element opposite the formula of the compound for the quantity of formula units and molecules shown:

For example:  $5\text{P}_2\text{O}_5$        $\text{P} = \underline{(5 \times 2 = ) 10}$        $\text{O} = \underline{(5 \times 5 = ) 25}$

For example:  $4\text{Zn}(\text{NO}_3)_2$        $\text{Zn} = \underline{(4 \times 1 = ) 4}$        $\text{N} = \underline{(4 \times 1 \times 2 = ) 8}$        $\text{O} = \underline{(4 \times 3 \times 2 = ) 24}$



# Answer Key For Counting Atoms In Compound

**Frederick A. Bettelheim,Jerry March**

## **Answer Key For Counting Atoms In Compound:**

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