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Chemistry 11

Classifying Matter

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

| | |
|------------|----------------|
| Elements | Carbon |
| Mixtures | Pure |
| Compound | Letters |
| Symbols | Carbon dioxide |
| Properties | |

- Elements cannot be separated by physical or chemical means.
- Elements are abbreviated with symbols, which consist of one or two letters.
- Two or more elements chemically combined make up a compound.
- Carbon is an example of an element and carbon dioxide is an example of a compound.
- Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

| Element Name | Element Symbol | Common Ions Formed |
|--------------|----------------|--------------------|
| a. Sodium | Na | +1 |
| b. Copper | Cu | +2, +1 |
| c. Tungsten | W | +6 |
| d. Iron | Fe | +3, +2 |
| e. Tin | Sn | +4, +2 |
| f. Potassium | K | +1 |
| g. Gold | Au | +3, +1 |
| h. Silver | Ag | +1 |

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

| | | | |
|--------------|---|--------------|---|
| a. P^{2-} | I | c. Ge | A |
| b. H_2O | M | d. O | A |
| c. O_2 | M | e. Ca^{2+} | I |
| d. Au^{3+} | I | g. NH_3 | M |

Answer Chemistry In The Lab

Emilie Sanchez

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