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Chemistry 11

Classifying Matter

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements	Carbon
Mixtures	Pure
Compound	Letters
Symbols	Carbon dioxide
Properties	

- Elements cannot be separated by physical or chemical means.
- Elements are abbreviated with symbols, which consist of one or two letters.
- Two or more elements chemically combined make up a compound.
- Carbon is an example of an element and carbon dioxide is an example of a compound.
- Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	Na	+1
b. Copper	Cu	+2, +1
c. Tungsten	W	+6
d. Iron	Fe	+3, +2
e. Tin	Sn	+4, +2
f. Potassium	K	+1
g. Gold	Au	+3, +1
h. Silver	Ag	+1

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a. P^{2-}	I	c. Ge	A
b. H_2O	M	d. O	A
c. O_2	M	e. Ca^{2+}	I
d. Au^{3+}	I	g. NH_3	M

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Christian Drosten

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1. Understanding the eBook Answer Key For Science Notebook
 - The Rise of Digital Reading Answer Key For Science Notebook
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Science Notebook
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Science Notebook
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Science Notebook
 - Personalized Recommendations
 - Answer Key For Science Notebook User Reviews and Ratings
 - Answer Key For Science Notebook and Bestseller Lists
5. Accessing Answer Key For Science Notebook Free and Paid eBooks
 - Answer Key For Science Notebook Public Domain eBooks
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 - Answer Key For Science Notebook Enhanced eBook Features
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- Adjustable Fonts and Text Sizes of Answer Key For Science Notebook
 - Highlighting and Note-Taking Answer Key For Science Notebook
 - Interactive Elements Answer Key For Science Notebook
8. Staying Engaged with Answer Key For Science Notebook
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Science Notebook
9. Balancing eBooks and Physical Books Answer Key For Science Notebook
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Science Notebook
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key For Science Notebook
- Setting Reading Goals Answer Key For Science Notebook
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key For Science Notebook
- Fact-Checking eBook Content of Answer Key For Science Notebook
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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