

# Ionic Compounds Practice

1. Name these compounds:



2. Write the chemical formula for each of the following:

(a) potassium iodide

(b) aluminum oxide

(c) calcium chloride

(d) zinc bromide

(e) sodium fluoride

(f) sodium oxide

3. Name these compounds:



4. Write the chemical formula for each of the following:

(a) calcium iodide

(b) calcium bromide

(c) barium carbonate

(d) calcium iodide

(e) calcium bromide

(f) calcium bromide

(g) calcium iodide

(h) calcium bromide

(i) manganese(IV) sulfide

(j) calcium iodide

(k) calcium bromide

(l) manganese(IV) sulfide

(m) calcium iodide

(n) calcium bromide

(o) manganese(IV) sulfide

(p) calcium iodide

(q) calcium bromide

(r) manganese(IV) sulfide

(s) calcium iodide

(t) calcium bromide

(u) manganese(IV) sulfide

# Answer Key For Making Models Of Compounds

**Lloyd S. Shapley**

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